

Total Synthesis of Oxazole- and Cyclopropane-Containing Epothilone A Analogues by the Olefin Metathesis Approach

K. C. Nicolaou,* Hans Vallberg, N. Paul King, Frank Roschangar, Yun He, Dionisios Vourloumis, and Christopher G. Nicolaou

Abstract: For structure–activity relationship studies, two series of epothilone A (**1**) analogues have been designed and synthesized, one containing an oxazole moiety instead of the thiazole heterocycle and the other containing a spirocyclopropane moiety in place of the *gem*-dimethyl group at position C-4 (4,4-ethano-epothilones). The olefin metathesis strategy in solution

was utilized for the chemical synthesis of these compounds starting with key building blocks **7–9** for the oxazole series (compounds **2**, **14–18**, **21–26**) and build-

ing blocks **8**, **30**, and **31** for the 4,4-ethano series (compounds **3**, **39–43**, **46–51**). The convergent strategy towards the designed epothilone A series involved a) an aldol condensation reaction, b) an esterification reaction, c) an olefin metathesis reaction catalyzed by $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$, and d) epoxidation of the macrocycle double bond.

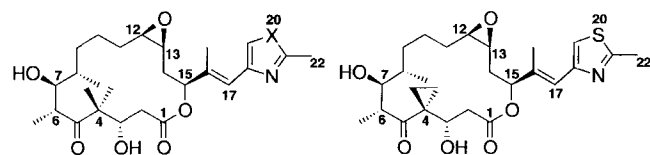
Keywords

epothilone · oxazoles · cyclopropanes · metathesis · total synthesis

Introduction

Amongst the most exciting antitumor agents of recent times are the naturally occurring epothilones A (**1**) (Figure 1) and B.^[1–2] Isolated^[1] from fermentation broths of myxobacteria *Sorangium cellulosum* strain 90 found in a soil sample originally collected from the banks of the Zambezi River in southern Africa, these substances exhibit potent antifungal and cytotoxic properties against a variety of cell lines.^[1–3] Of particular interest is their demonstrated ability to cause death in TaxolTM-resistant tumor cells and their Taxol-like mechanism of action, which involves tubulin assembly and stabilization of microtubules.^[4] Following the disclosure of their stereostructure by spectro-

scopic and X-ray crystallographic techniques in 1996,^[5] numerous synthetic studies^[6] and several total syntheses of epothilones A (**1**)^[7–11, 13] and B^[10, 12, 13] have been reported. In addition, several analogues of these compounds have already been synthesized and studied.^[6a, 7, 10–15] In this article we wish to describe the details of the total synthesis of a series of epothilone A analogues, in which either the thiazole has been replaced by an oxazole moiety or the *gem*-dimethyl group has been substituted with a cyclopropane ring. Minimized structures of (12*R*,13*S*)-epothilone A (**1**), (12*R*,13*S*)-20-oxa-epothilone A (**2**), and (12*R*,13*S*)-4,4-ethanoepothilone A (**3**) are shown in Figure 2. As expected, these modeling studies confirmed the strong conformational similarities between **1** and **2**, but revealed some differences with the 4,4-ethano analogue **3**. It was, therefore, of interest to synthesize these and related compounds for biological investigations in order to establish structure–activity relationships for drug design purposes. In the accompanying article,^[16] we describe an analogous series of epothilone B-related compounds.



1: X = S : epothilone A

2: X = O : 20-oxa-epothilone A

3: 4,4-ethano-epothilone A

Figure 1. Structure and numbering of epothilone A (**1**) and analogues **2** and **3**.

[*] Prof. Dr. K. C. Nicolaou, Dr. H. Vallberg, Dr. N. P. King, Dr. F. Roschangar, Y. He, Dr. D. Vourloumis, C. G. Nicolaou
Department of Chemistry and The Skaggs Institute for Chemical Biology
The Scripps Research Institute, 10550 North Torrey Pines Road
La Jolla, California 92037 (USA)
and
Department of Chemistry and Biochemistry,
University of California, San Diego
9500 Gilman Drive, La Jolla, California 92093 (USA)
Fax: Int. code + (619) 784-2469

Results and Discussion

The oxazole series of epothilone A analogues: Based on our original olefin metathesis approach to epothilones published in 1996,^[6a] we undertook the synthesis of an oxazole series of epothilone A analogues, represented by structure **2**, outlined retrosynthetically in Scheme 1. Thus, building blocks **8**, **9**, and **10** were to be assembled and converted to the more advanced intermediates **6** and **7** by means of an aldol condensation and an asymmetric allylboration reaction, respectively. The latter compounds were then to be joined by an esterification and the cou-

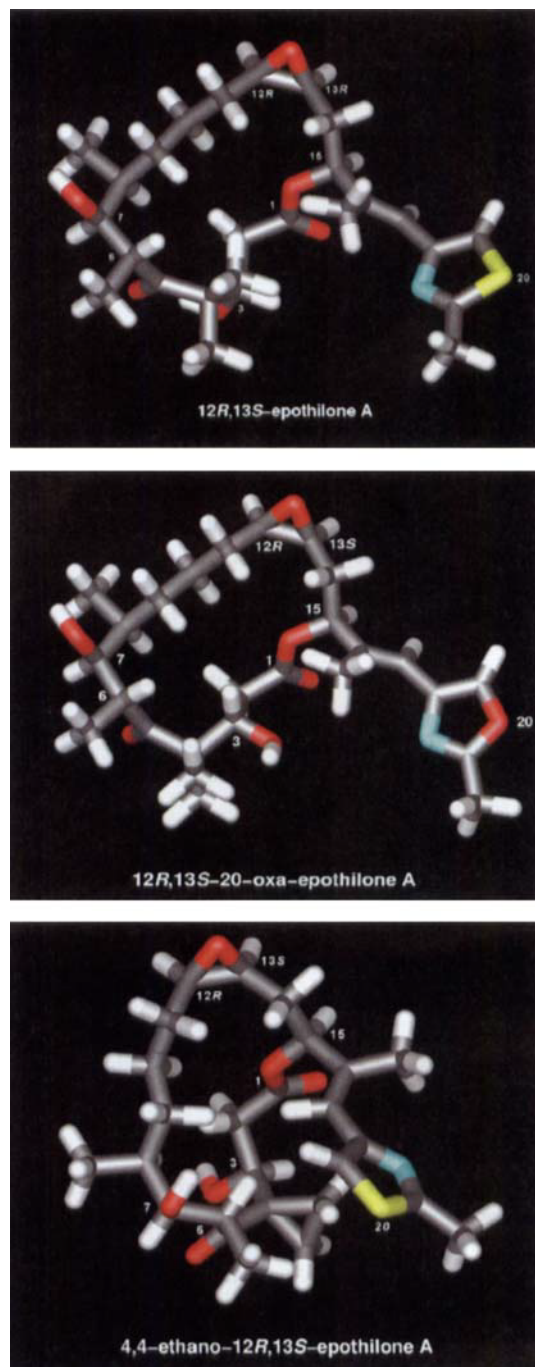
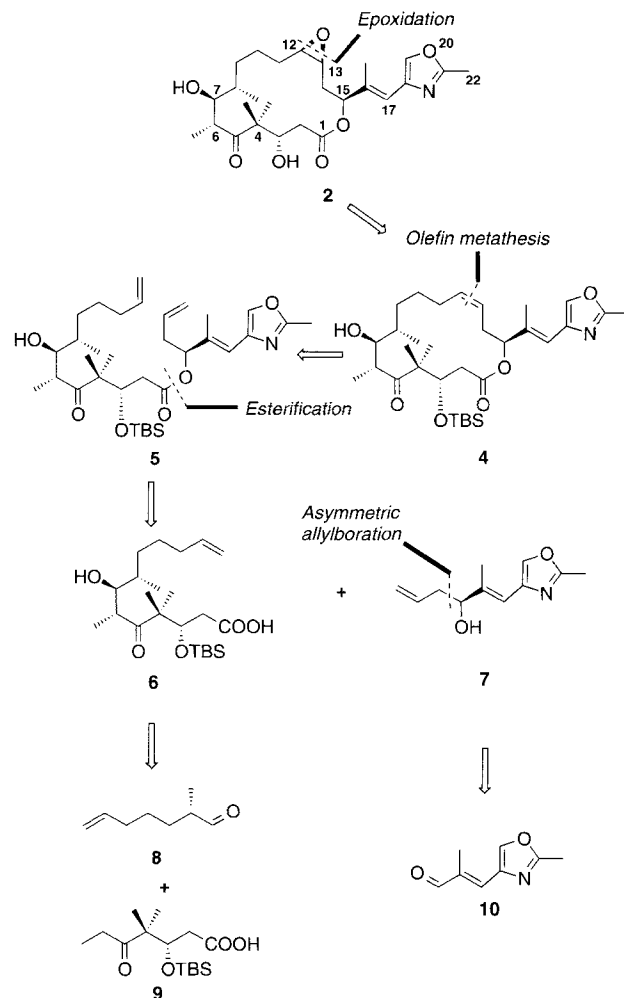


Figure 2. Computer-generated, minimized structures of (12*R*,13*S*)-epothilone A (**1**, top), (12*R*,13*S*)-20-oxa-epothilone A (**2**, center), and 4,4-ethano-(12*R*,13*S*)-epothilone A (**3**, bottom). The epothilone atoms are colored according to the following code: carbon, grey; hydrogen, white; oxygen, red; nitrogen, blue; sulfur, yellow.

pled product cyclized by an olefin metathesis reaction leading to olefinic precursor **4**. Finally, deprotection and epoxidation of **4** was expected to give the desired target molecule **2**. The expected lack of stereoselectivity at the aldol and olefin metathesis stages of the sequence was a desirable feature, since the immediate goal of our research program was to generate as diverse a library of compounds as possible for biological screening.

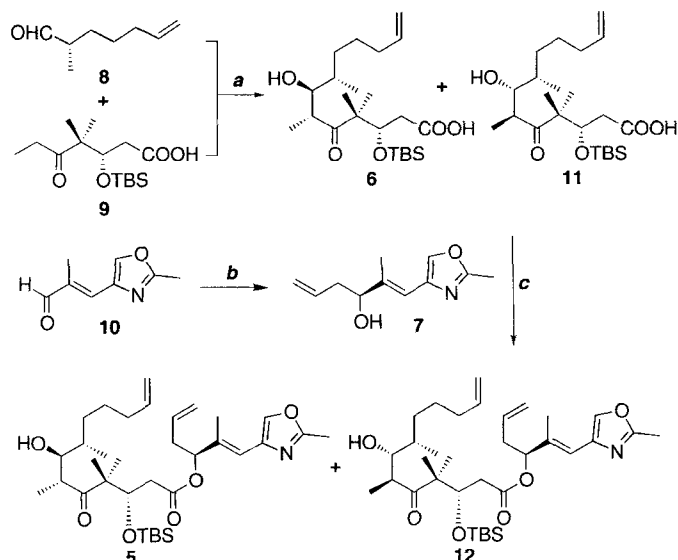
Scheme 2 summarizes the construction of olefin metathesis precursors **5** and **12**. Thus, condensation of aldehyde **8**^[11] with the dianion of **9**^[13] generated by the action of 2.4 equiv of LDA



Scheme 1. Retrosynthetic analysis of 20-oxa-epothilone A analogue **2**.

(for abbreviations, see legends in schemes) in THF at -40°C , yielded aldols **6** and **11** in a ratio of approximately 5:3. Reaction of the known oxazole derivative **10**^[17] with (+)-Ipc₂B(allyl) in ether at -100°C ^[18] furnished allylic alcohol **7** in 92% yield and $\geq 95\%$ ee (as determined by formation of the Mosher ester).^[19]

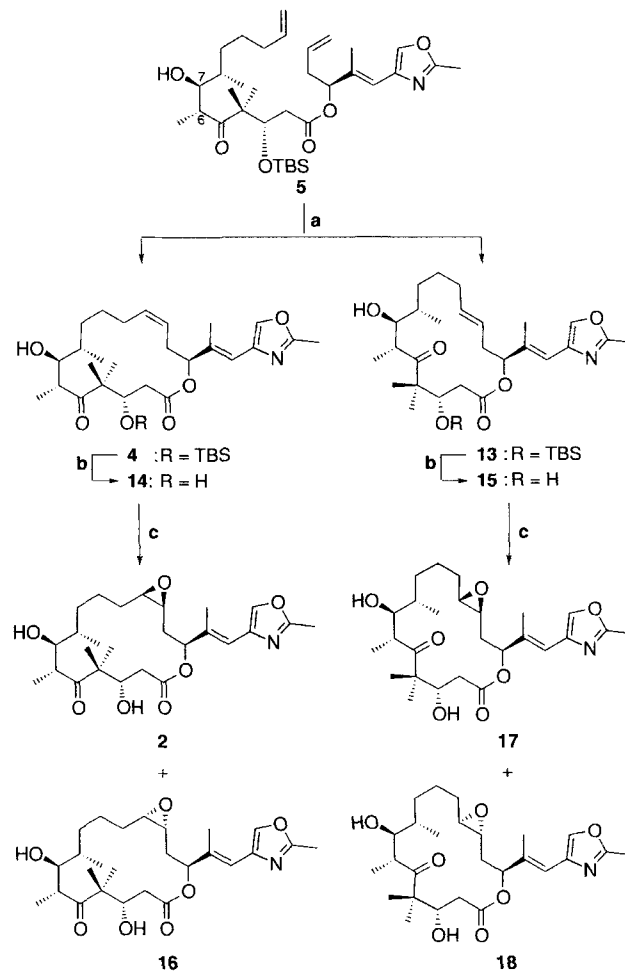
Abstract in Greek: Δοο καινουργίες ρειρές αναλογών της εποθειλονής Α (**1**) ρχεδιαστηκαν και συντεθηκαν αποσκοποντας στην μελετη της ρχεσης μεταξύ δομης και βιολογικης δρασης: η πρώτη περιχει μια οξάζολη αντι του ετεροκυκλικου δακτηλιου της θειάζολης, ενώ η δεύτερη περιχει μια ρπιρο-εθανικη ομάδα αντι των δύο μεθυλιων στη θέση 4 (4,4-εθανο-εποθειλονες). Η στρατηγικη της ολεφινικης μεταθεσης χρησιμοποιηθηκε για τη ρνθεση αυτων των ουσιων ξεκινωντας απο τις δομικες ουσιες 7–9 για τη ρειρα οξάζολης (ουσιες **2**, **14–18**, **21–26**) και τις δομικες ουσιες **8**, **30** και **31** για τη ρειρα 4,4-εθανο (ουσιες **3**, **39–43**, **46–51**). Η ελικτικη στρατηγικη που ακολουθειται για τη ρνθεση των αναλογων αυτων της εποθειλονής Α αποτελειται απο: α) μια αντιδραση αλδοολικης ρυμπυκνωσης, β) μια αντιδραση εστεροποιησης, γ) μια καταλυτικη αντιδραση ολεφινικης μεταθεσης παρουνια [RuCl₂(=CHPh)(PCy₃)₂], και τελος δ) εποξειδωση του μακροκυκλικου διπλου δερμου.



Scheme 2. Coupling of building blocks and construction of precursors **5** and **12**. Reagents and conditions: a) 2.2 equiv of LDA, -40°C , THF, 1.5 h, then **8** in THF, -40°C , 0.5 h (**6**:**11** ca. 5:3); b) 1.2 equiv of (+)-Ipc,B(allyl), Et_2O , -100°C , 0.5 h, 92%; c) 2.5 equiv of **7**, 3.0 equiv of EDC, 0.1 equiv of 4-DMAP, CH_2Cl_2 , $0 \rightarrow 25^{\circ}\text{C}$, 12 h, 44% (**5**) plus 28% (**12**) for two steps. TBS = *tert*-butyldimethylsilyl; Ipc = isopinocampheyl, LDA = lithium diisopropylamide; EDC = 1-ethyl-3-(dimethylaminopropyl)-3-carbodiimide hydrochloride; 4-DMAP = 4-dimethylaminopyridine.

Coupling of **7** with the mixture of **6** and **11**, in the presence of EDC and 4-DMAP, followed by chromatographic separation furnished pure **5** (44% yield from **8** + **9**) and **12** (28% yield from **8** + **9**).

The olefin metathesis of **5** (Scheme 3) was facilitated by $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$ catalyst (Cy = cyclohexyl)^[20–22] and led to a mixture of *cis* and *trans* cyclic olefins. Chromatographic purification furnished pure compounds **4** (40% yield) and **13** (32% yield), which were processed separately. Thus, exposure of **4** to 20% TFA in CH_2Cl_2 at ambient temperature gave diol **14** (89% yield), and similar treatment of **13** furnished **15** (95% yield). Reaction of **14** with methylperoxycarboximide acid $[\text{CH}_3\text{C}(=\text{NH})\text{OOH}$, prepared in situ from acetonitrile and 35% aq. H_2O_2 in the presence of KHCO_3]^[23] resulted in the formation of epoxide **2** (52% yield based on ca. 50% conversion) along with a trace amount of **16** (ratio of **2**:**16** > 20:1 by ^1H NMR), whereas similar reaction of **15** led to **17** (17% yield) and **18** (24% yield) (based on ca. 50% conversion).^[24] These epoxides were purified by preparative thin-layer chromatography, and their stereochemistry was assigned by comparison of their NMR spectra with the original epothilone A series, the stereochemistry of which was determined by ^1H – ^1H NOESY and ^1H – ^1H COSY experiments and molecular modeling,^[11] and in accordance with the higher potencies of **2** and **17** in tubulin polymerization experiments as compared to those of **16** and **18**, respectively.^[25] However, it must be emphasized that these stereochemical assignments are tentative, requiring further experimental evidence. It is worth noting that application of these epoxidation conditions to the actual epothilone A (**1**) olefinic precursor resulted in improved diastereoselectivity (ca. 13:1 in favor of the desired isomer, 65% combined yield based on ca. 75% conversion) over our previously reported methods.^[11]

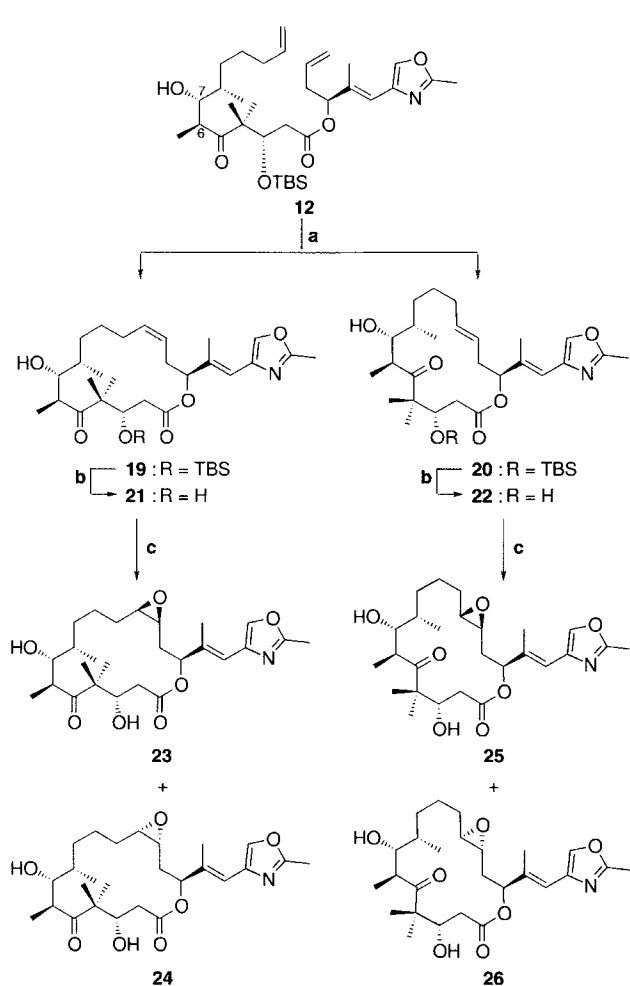


Scheme 3. Olefin metathesis of precursor **5** and synthesis of 20-oxa-epothilone A analogues **2**, **16**–**18**. Reagents and conditions: a) 20 mol% of $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$, CH_2Cl_2 , 25°C , 16 h, 40% (**4**) plus 32% (**13**); b) 20% TFA in CH_2Cl_2 , 25°C , 2 h, 89% (**14**), 95% (**15**); c) 35% H_2O_2 , CH_3CN , KHCO_3 , MeOH, 25°C , 52% (**2**) plus trace of **16**; 17% (**17**) plus 24% (**18**) (all yields based on ca. 50% conversion). TFA = trifluoroacetic acid. The tentative stereochemical assignments of epoxides **2** and **17** were based on NMR comparisons with members of the natural epothilone A series as well as the higher potencies of **2** and **17** in the tubulin polymerization assay as compared to those of **16** and **18**, respectively.

Processing of the (6*S*,7*R*) diastereomeric aldol product **12** in the same way as described above for **5**, led to the oxazole-containing epothilone A analogues **19**–**26**, as summarized in Scheme 4. The tentative stereochemical assignments of epoxides **23**–**26** were based on NMR correlations to the corresponding epothilone A analogues.

The 4,4-ethano series of epothilone A analogues: Following the same retrosynthetic rationale as the one outlined above for the oxazole analogues, the 4,4-ethano-epothilone A was analyzed as shown in Scheme 5. This time, the analysis led to building blocks **30**, **8**, and **31**. Compound **31** was further broken down into β -ketoester **34** via intermediates **32** and **33**.

We began with the synthesis of cyclopropylketoacid **31** (Scheme 6). Thus, reaction of 1,2-dibromoethane with ethyl propionylacetate (**34**) in the presence of K_2CO_3 at ambient temperature resulted in the formation of cyclopropylketoester **35** (60% yield).^[26] Reduction of the ester and keto groups with LiAlH_4 (93% yield) followed by Swern oxidation of the result-

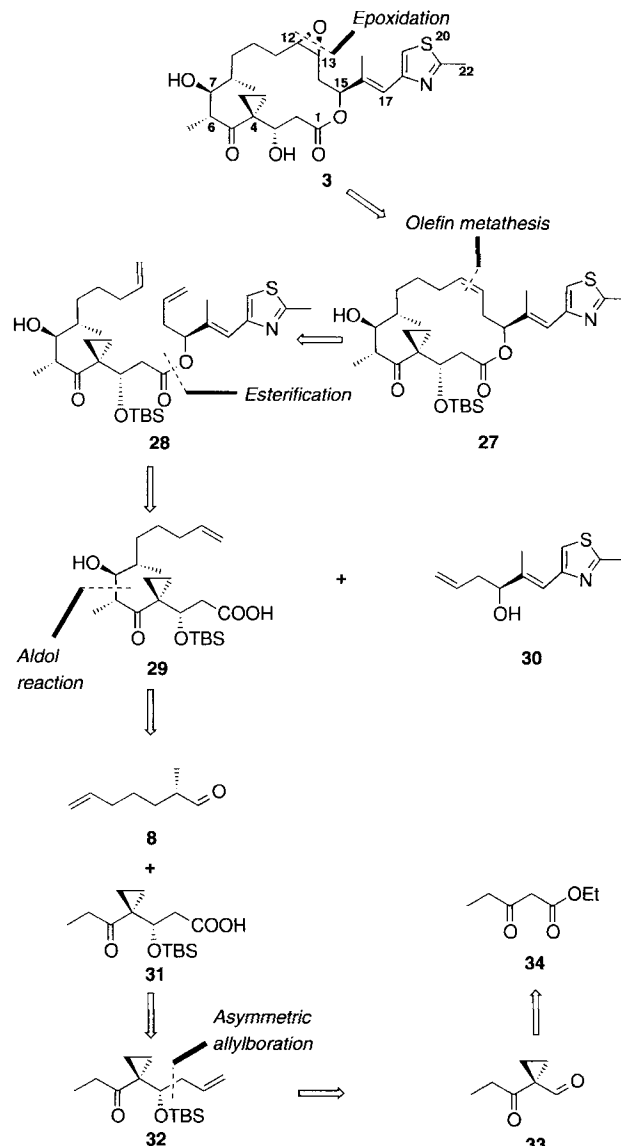


Scheme 4. Olefin metathesis of C6–C7 diastereomeric precursor **12** and synthesis of 20-oxa-epothilone A analogues **23–26**. Reagents and conditions: a) 20 mol% of $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$, CH_2Cl_2 , 25 °C, 20 h, 25% (**19**) plus 63% (**20**); b) 20% TFA in CH_2Cl_2 , 25 °C, 2 h, 75% (**21**), 72% (**22**); c) 35% H_2O_2 , CH_3CN , KHCO_3 , MeOH, 25 °C, 24% (**23** or **24**) plus 9% (**24** or **23**), 15% (**25** or **26**) plus 19% (**26** or **25**).

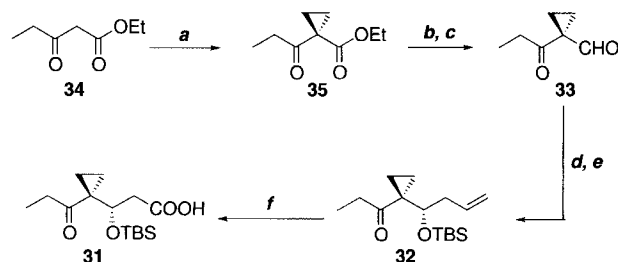
ing diol $[(\text{COCl})_2; \text{DMSO}; \text{Et}_3\text{N}]$ furnished ketoaldehyde **33** in 64% yield. Chemo- and stereoselective addition of (+)- $\text{Ipc}_2\text{B}(\text{allyl})^{[18]}$ to aldehyde **33** (> 85% *ee* by Mosher ester analysis),^[19] followed by silylation (TBSOTf; 2,6-lutidine) of the generated secondary alcohol, gave silyl ether **32**. Finally, cleavage of the terminal olefin in **32** with NaIO_4 in the presence of catalytic amounts of $\text{RuCl}_3 \cdot \text{H}_2\text{O}$ in $\text{MeCN}/\text{H}_2\text{O}/\text{CCl}_4$ (2:3:2) at 25 °C^[27] yielded the desired ketoacid **31** in 43% overall yield from cyclopropylketoaldehyde **33**.

The dianion of ketoacid **31** (LDA in THF at -30 °C) reacted with aldehyde **8** to form aldols **29** and **36** in a ratio of approximately 2:3 (determined by $^1\text{H NMR}$) (Scheme 7). The coupling of the mixture of **29** and **36** with fragment **30**^[31] was facilitated by EDC and 4-DMAP, and the resulting hydroxyesters were chromatographically separated to afford pure **28** (15%) and **37** (36%).

Ring closure of advanced intermediate **28** and epoxidation of the desilylated cyclic diols are shown in Scheme 8. Thus, stirring of **28** with $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$ catalyst in CH_2Cl_2 at 25 °C followed by chromatographic separation (silica gel, preparative thin layer) furnished *cis* and *trans* olefins **27** (37% yield) and **38**

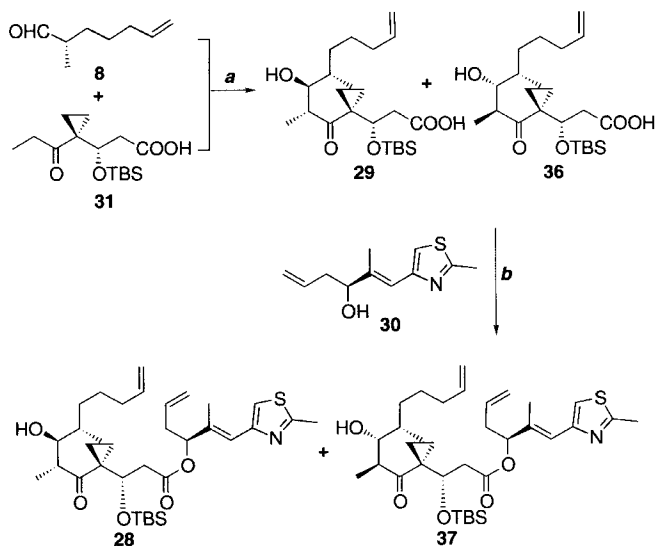


Scheme 5. Retrosynthetic analysis of 4,4-ethano-epothilone A analogue **3**.

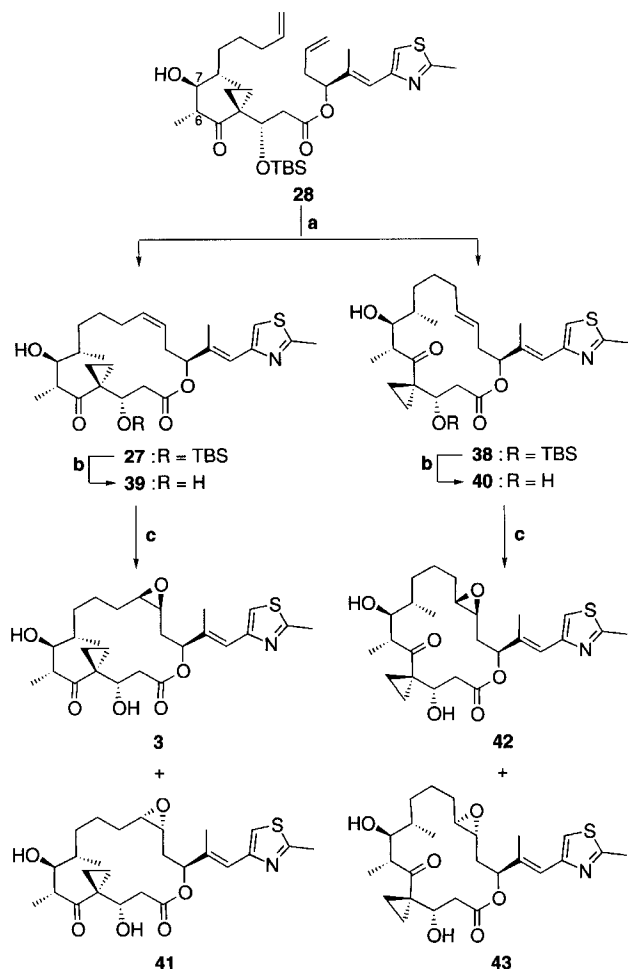


Scheme 6. Synthesis of ketoacid **31**. Reagents and conditions: a) 1.3 equiv of $\text{BrCH}_2\text{CH}_2\text{Br}$, 3.0 equiv of K_2CO_3 , DMF, 25 °C, 15 h, 60%; b) 2.0 equiv of LiAlH_4 , Et_2O , $-20 \rightarrow 0$ °C, 2.5 h, 93%; c) 4.0 equiv of DMSO, 3.0 equiv of $(\text{COCl})_2$, 8.0 equiv of Et_3N , CH_2Cl_2 , $-78 \rightarrow 0$ °C, 64%; d) 1.1 equiv of (+)- $\text{Ipc}_2\text{B}(\text{allyl})$, Et_2O , -100 °C; e) 3.8 equiv of TBSOTf, 4.6 equiv of 2,6-lutidine, CH_2Cl_2 , -78 °C; f) 4.1 equiv of NaIO_4 , 0.05 equiv of $\text{RuCl}_3 \cdot \text{H}_2\text{O}$, $\text{MeCN}:\text{H}_2\text{O}:\text{CCl}_4$ (2:3:2), 25 °C, 43% for three steps. DMSO = dimethyl sulfoxide; TBS = *tert*-butyldimethylsilyl; Ipc = isopinocampheyl.

(35% yield), respectively. The corresponding diols **39** (65% yield) and **40** (62% yield) were obtained by treating the respective silyl ethers with 25% HF·pyridine in THF at ambient temperature. Finally, epoxidation of **39** with methyl(trifluoro-



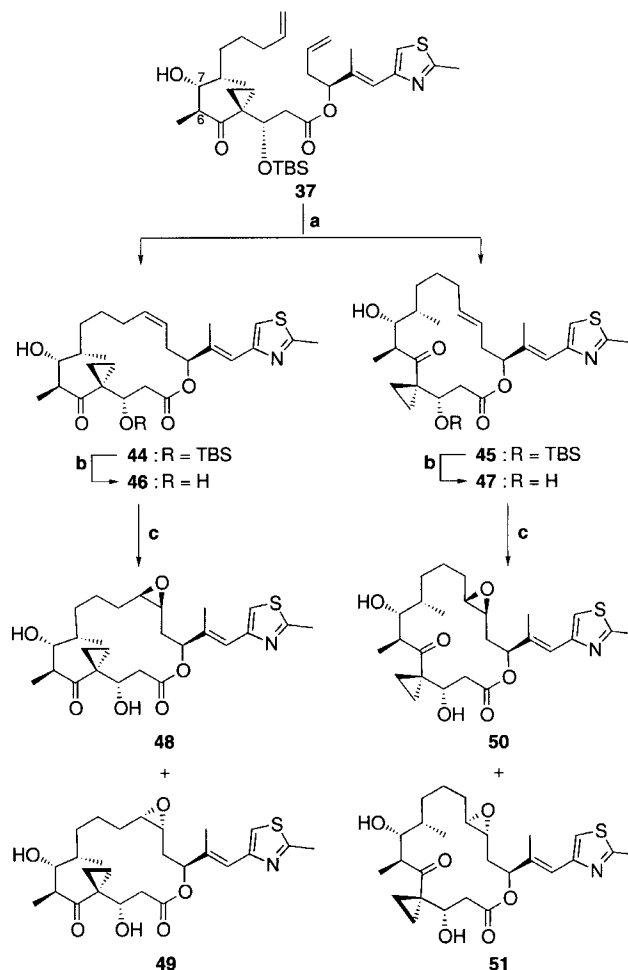
Scheme 7. Coupling of building blocks and construction of advanced intermediates **28** and **37**. Reagents and conditions: a) 2.4 equiv of LDA, -30°C , THF, 2 h, then **8** in THF, -30°C , 0.5 h, (**29**:**36** ca. 2:3); c) 2.5 equiv of **30**, 1.2 equiv of EDC, 0.1 equiv of 4-DMAP, CH_2Cl_2 , $0 \rightarrow 25^{\circ}\text{C}$, 2 h, 15% (**28**) plus 36% (**37**) for two steps. TBS = *tert*-butyldimethylsilyl; LDA = lithium diisopropylamide; EDC = 1-ethyl-(3-dimethylaminopropyl)-3-carbodiimide hydrochloride; 4-DMAP = 4-dimethylaminopyridine.



Scheme 8. Olefin metathesis of diene **28** and synthesis of 4,4-ethano-epothilone A analogues **3** and **41–43**. Reagents and conditions: a) 10 mol% of $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$, CH_2Cl_2 , 25°C , 2 h, 37% (**27**) plus 35% (**38**); b) 25% HF·Py in THF, $0 \rightarrow 25^{\circ}\text{C}$, 28 h, 65% (**39**), 62% (**40**); c) CH_2Cl_2 : CH_3CN : Na_2EDTA (1:2:1.5), 50 equiv of CF_3COCH_3 , 11 equiv of NaHCO_3 , 7.0 equiv of Oxone[®], 0°C , 50% (**3** or **41**) plus 29% (**41** or **3**); 11% (**42** or **43**) plus 31% (**43** or **42**).

methyl)dioxirane^[10,28] gave epoxides **3** and **41** (stereochemistries not assigned, 50 and 29% yields), whereas similar treatment of **40** furnished **42** and **43** (stereochemistries not assigned, 11 and 31% yields).

The other aldol product, compound **37**, was processed in a similar way as described above for **28**, furnishing the 4,4-ethano-epothilone A analogues **46–51** as shown in Scheme 9. Again, the stereochemical details of these compounds remain unassigned.



Scheme 9. Olefin metathesis of C6–C7 diastereomeric diene **37** and synthesis of 4,4-ethano-epothilone A analogues **48–51**. Reagents and conditions: a) 9 mol% of $[\text{RuCl}_2(=\text{CHPh})(\text{PCy}_3)_2]$, CH_2Cl_2 , 25°C , 1 h, 18% (**44**) plus 58% (**45**); b) 25% HF·Py in THF, $0 \rightarrow 25^{\circ}\text{C}$, 22 h, 54% (**46**), 76% (**47**); c) CH_2Cl_2 : CH_3CN : Na_2EDTA (4:4:1), 50 equiv of CF_3COCH_3 , 16 equiv of NaHCO_3 , 10 equiv of Oxone[®], 0°C , 39% (**48** or **49**) plus 35% (**49** or **48**); 22% (**50** or **51**) plus 27% (**51** or **50**).

Conclusion

Applying the olefin metathesis approach to epothilones, we have synthesized a series of oxazole- and cyclopropane-containing epothilone A analogues. These compounds considerably enrich the known epothilone libraries in terms of molecular diversity and numbers. Biological investigations with these analogues established^[25] useful structure–activity relationships within this important class of compounds. Interestingly, while the oxazole series of compounds exhibited comparable tubulin polymerization activity and cytotoxicity to the corresponding thia-

zole series, the 4,4-ethano-epothilones proved inactive.^[25] These results underscore the importance of conformational precision in these compounds for biological action. In the accompanying article^[16] we describe the design and chemical synthesis of a corresponding series of epothilone B analogues with oxazole and cyclopropyl moieties utilizing the macrolactonization approach.

Experimental Section

General Techniques: All reactions were carried out under an argon atmosphere with dry, freshly distilled solvents under anhydrous conditions, unless otherwise noted. Tetrahydrofuran (THF), toluene, and diethyl ether (ether) were distilled from sodium benzophenone, and methylene chloride (CH_2Cl_2) from calcium hydride. Anhydrous solvents were also obtained by passing them through commercially available alumina column. Yields refer to chromatographically and spectroscopically (^1H NMR) homogeneous materials, unless otherwise stated. Reagents were purchased at highest commercial quality and used without further purification unless otherwise stated. Reactions were monitored by thin-layer chromatography carried out on 0.25 mm E. Merck silica gel plates (60 F-254) using UV light as visualizing agent and 7% ethanolic phosphomolybdic acid or *p*-anisaldehyde solution and heat as developing agents. E. Merck silica gel (60, particle size 0.040–0.063 mm) was used for flash column chromatography. Preparative thin-layer chromatography (PTLC) separations were carried out on 0.25, 0.50, or 1 mm E. Merck silica gel plates (60 F-254). NMR spectra were recorded on Bruker DRX-600, AMX-500, AMX-400, or AC-250 instruments and calibrated with residual undeuterated solvent as an internal reference. The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad. IR spectra were recorded on a Perkin-Elmer 1600 series FT-IR spectrometer. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter. High-resolution mass spectra (HRMS) were recorded on a VG ZAB-ZSE mass spectrometer under fast atom bombardment (FAB) conditions with NBA as the matrix. Melting points (m.p.) are uncorrected and were recorded on a Thomas Hoover Unimelt capillary melting point apparatus.

Hydroxyacids 6 and 11—aldol condensation of acid 9 with aldehyde 8: A solution of ketoacid **9** (1.131 g, 3.75 mmol, 1.0 equiv) in THF (30 mL) was added dropwise at -78°C to a freshly prepared solution of LDA [formed by addition of *n*BuLi (5.27 mL, 1.6 M solution in hexanes, 8.44 mmol, 2.25 equiv) to a solution of diisopropylamine (1.16 mL, 8.25 mmol, 2.2 equiv) in THF (30 mL) at -10°C and stirring for 30 min]. After stirring at the same temperature for 15 min, the reaction mixture was allowed to warm to -30°C and stirred at that temperature for 1.5 h. The reaction mixture was cooled back to -78°C and a solution of aldehyde **8** (0.757 g, 6.00 mmol, 1.6 equiv) in THF (15 mL) was added through a cannula. The resulting mixture was stirred for 15 min at -78°C , then warmed to -40°C and stirred for 1 h, cooled to -78°C , and quenched by slow addition of saturated aqueous NH_4Cl (50 mL) solution. The reaction mixture was warmed to 0°C , and a solution of aqueous 4N HCl (1.9 mL, 7.50 mmol, 2.0 equiv) was added, followed by warming to 25°C . Then the pH was adjusted to $\approx 2-3$ by dropwise addition of aqueous 4N HCl. Extractions with EtOAc (6 \times 25 mL), filtration through a short plug of silica gel, and concentration afforded, in high yield, a mixture of aldol products **6** and **11** along with unreacted starting acid **9** in a 56:36:8 ratio (^1H NMR). This crude material was used without further purification. $R_f = 0.20$ (silica gel, 50% EtOAc in hexanes); ^1H NMR (500 MHz, CDCl_3); only signals for **6** and **11** are reported: $\delta = 5.88-5.73$ (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.04–4.92 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.51–4.47 (m, 0.4H, CHOTBS), 4.44–4.40 (m, 0.6H, $(\text{CH}_3)_2\text{CCH}(\text{OTBS})$), 3.42 (d, $J = 8.0$ Hz, 0.4H, $\text{CHOH}(\text{CHCH}_3)$), 3.32 (d, $J = 9.0$ Hz, 0.6H, $\text{CHOH}(\text{CHCH}_3)$), 3.30–3.20 (m, 1H, $\text{CH}_2\text{CH}(\text{C}=\text{O})$), 2.51–2.45 (m, 1H, CH_2COOH), 2.38 (dd, $J = 16.5, 6.5$ Hz, 0.4H, CH_2COOH), 2.35 (dd, $J = 16.5, 6.5$ Hz, 0.6H, CH_2COOH), 2.11–1.98 (m, 2H), 1.80–1.21 (m, 5H), 1.20 (s, 1.8H, $\text{C}(\text{CH}_3)_2$), 1.19 (s, 1.2H, $\text{C}(\text{CH}_3)_3$), 1.16 (s, 1.8H, $\text{C}(\text{CH}_3)_2$), 1.14 (s, 1.2H, $\text{C}(\text{CH}_3)_3$), 1.06 (d, $J = 6.5$ Hz, 1.2H), 1.05 (d, $J = 6.5$ Hz, 1.8H), 1.00 (d, $J = 6.5$ Hz, 1.2H), 0.89 (s, 5.4H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.87 (s, 3.6H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.85 (d, $J = 7.0$ Hz, 1.8H), 0.11 (s, 1.8H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.09 (s, 1.2H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.08 (s, 1.2H,

$\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.07 (s, 1.8H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$); HRMS (FAB): calcd for $\text{C}_{23}\text{H}_{44}\text{NaO}_5\text{Si}$ ($M + \text{Na}^+$) 451.2856, found 451.2867.

Alcohol 7—allylation of aldehyde 10: Aldehyde **10** (24.6 g, 163 mmol) was dissolved in anhydrous ether (550 mL, 0.3 M) and the solution was cooled to -100°C . (+)-Allyldiisopinocampheylborane (0.15 M in pentane; 1.3 L, 196 mmol, 1.2 equiv; prepared from (–)-Ipc₂BOMe and 1.0 equiv of allylmagnesium bromide) was added dropwise under vigorous stirring, and the reaction mixture was allowed to stir for 1 h at the same temperature. Methanol was added at -100°C , and the reaction mixture was allowed to warm to room temperature. Amino ethanol (50 mL, 0.81 mol, 5.0 equiv) was added, and stirring was continued for 15 h. The workup procedure was completed by the addition of saturated aqueous NH_4Cl solution, extraction with EtOAc and drying of the combined organic layers with MgSO_4 . Filtration, followed by evaporation of the solvents under reduced pressure and flash column chromatography (silica gel, 35% ether in hexanes until all the boron complexes were removed; then 70% ether in hexanes) provided alcohol **7** (28.8 g, 92%). $R_f = 0.41$ (silica gel, 33% EtOAc in hexanes); $[\alpha]_D^{25} = -22.2$ ($c = 1.10$, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}} = 3342, 2977, 2930, 2858, 1584, 1107$ cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): $\delta = 7.48$ (s, 1H, ArH), 6.30 (s, 1H, ArCH=CCH₃), 5.84–5.76 (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.17–5.12 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.21 (dd, $J = 7.5, 5.0$ Hz, 1H, CHOH), 2.46 (s, 3H, CH_3 Ar), 2.46–2.31 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 1.94 (s, 3H, ArCH=CCH₃); ^{13}C NMR (125.7 MHz, CDCl_3): $\delta = 160.7, 141.3, 137.8, 135.3, 134.5, 117.8, 115.2, 76.1, 39.9, 14.6, 13.7$; HRMS (FAB): calcd for $\text{C}_{11}\text{H}_{16}\text{NO}_2$ ($M + \text{H}^+$) 194.1181, found 194.1186.

Esters 5 and 12—EDC coupling of alcohol 7 with the mixture of ketoacids 6 and 11: A solution of ketoacids **6** and **11** (981 mg, 2.30 mmol, 1.0 equiv), 4-(dimethylamino)pyridine (4-DMAP, 28 mg, 0.23 mmol, 0.1 equiv), and alcohol **7** (1.11 g, 5.74 mmol, 2.5 equiv) in CH_2Cl_2 (1.2 mL, 2 M) was cooled to 0°C and then treated with 1-ethyl-(3-dimethylaminopropyl)-3-carbodiimide hydrochloride (EDC, 1.32 g, 6.90 mmol, 3.0 equiv). The reaction mixture was stirred at 0°C for 2 h and then at 25°C for 12 h. The solution was concentrated to dryness in vacuo, and the residue was taken up in EtOAc (10 mL) and water (10 mL). The organic layer was separated, washed with saturated NH_4Cl solution (10 mL) and water (10 mL), and dried (MgSO_4). Evaporation of the solvents followed by flash column chromatography (silica gel, 15% EtOAc in hexanes) resulted in the isolation of the pure hydroxyesters **5** (0.608 g, 44% from ketoacid **9**) and **12** (0.387 g, 28% from ketoacid **9**).

5: $R_f = 0.70$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{25} = -36.6$ ($c = 1.31$, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}} = 3508, 2945, 2857, 1737, 1685, 1587, 1095$ cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): $\delta = 7.43$ (s, 1H, ArH), 6.21 (s, 1H, ArCH=CCH₃), 5.80–5.72 (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.70–5.62 (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.25 (dd, $J = 7.0, 6.5$ Hz, 1H, CO_2CH), 5.06 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.01 (d, $J = 10.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.95 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.88 (d, $J = 10.5$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.37 (dd, $J = 6.0, 3.5$ Hz, 1H, $(\text{CH}_3)_2\text{CCH}(\text{OTBS})$), 3.44 (s, 1H, $\text{CHOH}(\text{CHCH}_3)$), 3.27–3.22 (m, 2H, $\text{CH}_2\text{CH}(\text{C}=\text{O})$ and $\text{CHOH}(\text{CHCH}_3)$), 2.45–2.38 (m, 3H), 2.40 (s, 3H, CH_3 Ar), 2.29 (dd, $J = 17.5, 6.0$ Hz, 1H, CH_2COO), 2.02–1.90 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 1.91 (s, 3H, ArCH=CCH₃), 1.74–1.67 (m, 1H), 1.52–1.36 (m, 2H), 1.29–0.95 (m, 2H), 1.16 (s, 3H, $\text{C}(\text{CH}_3)_2$), 1.07 (s, 3H, $\text{C}(\text{CH}_3)_3$), 1.00 (d, $J = 7.0$ Hz, 3H, $\text{CH}_3\text{CH}(\text{C}=\text{O})$), 0.85 (s, 9H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.79 (d, $J = 7.0$ Hz, 3H, CH_3CHCH_2), 0.08 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.02 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$); ^{13}C NMR (125.7 MHz, CDCl_3): $\delta = 221.7, 170.8, 160.7, 139.0, 137.5, 136.3, 135.8, 133.1, 117.8, 117.3, 114.1, 78.4, 74.5, 73.3, 53.8, 41.2, 40.1, 37.4, 35.4, 34.1, 32.3, 26.0, 25.9, 21.9, 19.8, 18.1, 15.2, 14.8, 13.7, 9.7, -4.4, -4.9$; HRMS (FAB): calcd for $\text{C}_{34}\text{H}_{57}\text{CsNO}_6\text{Si}$ ($M + \text{Cs}^+$) 736.3010, found 736.3036.

12: $R_f = 0.63$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{25} = -11.0$ ($c = 0.77$, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}} = 3501, 2931, 2872, 1738, 1692, 1587, 1090$ cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): $\delta = 7.44$ (s, 1H, ArH), 6.21 (s, 1H, ArCH=CCH₃), 5.81–5.72 (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.71–5.62 (m, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.28 (dd, $J = 7.0, 7.0$ Hz, 1H, CO_2CH), 5.06 (d, $J = 17.5$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.01 (d, $J = 11.5$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.98 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.92 (d, $J = 10.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.46 (dd, $J = 6.0, 4.0$ Hz, 1H, $(\text{CH}_3)_2\text{CCH}(\text{OTBS})$), 3.44–3.38 (m, 2H, $\text{CHOH}(\text{CHCH}_3)$ and $\text{CHOH}(\text{CHCH}_3)$), 3.19 (qd, $J = 7.0, 1.5$ Hz, 1H, $\text{CH}_2\text{CH}(\text{C}=\text{O})$), 2.48–2.39 (m, 3H), 2.40 (s, 3H, CH_3 Ar), 2.33 (dd, $J = 17.0, 6.0$ Hz, 1H, CH_2COO), 2.06–1.92 (m, 2H), 1.92 (s, 3H, ArCH=CCH₃), 1.51–0.95 (m, 5H), 1.12 (s, 3H, $\text{C}(\text{CH}_3)_2$), 1.08 (s, 3H, $\text{C}(\text{CH}_3)_3$), 0.99 (d, $J = 7.0$ Hz, 3H, $\text{CH}_3\text{CH}(\text{C}=\text{O})$), 0.96 (d, $J = 7.0$ Hz, 3H, CH_3CHCH_2), 0.83

(s, 9H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), 0.05 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), 0.03 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 220.7, 170.9, 160.7, 138.6, 137.6, 136.4, 135.8, 133.1, 117.8, 117.3, 114.5, 78.4, 74.8, 72.5, 53.9, 41.4, 40.1, 37.4, 35.3, 33.8, 32.1, 26.0, 25.9, 21.7, 19.6, 18.1, 15.4, 14.8, 13.7, 10.6, -4.4, -4.8; HRMS (FAB): calcd for $\text{C}_{34}\text{H}_{57}\text{CsNO}_6\text{Si}$ ($M + \text{Cs}^+$) 736.3010, found 736.3035.

Hydroxylactones 4 and 13—cyclization of diene 5 by olefin metathesis: To a solution of diene 5 (145 mg, 0.24 mmol) in CH_2Cl_2 (240 mL, 0.001 M) was added $[\text{RuCl}_2(\text{C}=\text{CHPh})(\text{PCy}_3)_2]$ (40 mg, 0.048 mol, 0.2 equiv), and the reaction mixture was allowed to stir at 25 °C for 16 h. After completion of the reaction (established by TLC), the solvent was removed under reduced pressure, and the crude products were purified by flash chromatography (silica gel, 20% EtOAc in hexanes) to give *cis*-hydroxylactone 4 (55 mg, 40%) and *trans*-hydroxylactone 13 (44 mg, 32%).

4: R_f = 0.19 (silica gel, 30% EtOAc in hexanes); $[\alpha]_D^{22}$ = -37.5 (c = 1.41, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3449, 2930, 2857, 1741, 1694, 1585, 1099, 733 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.47 (s, 1H, ArH), 6.27 (s, 1H, ArCH=CCH₃), 5.44 (ddd, J = 10.5, 10.5, 2.5 Hz, 1H, CH=CHCH₂), 5.33 (ddd, J = 10.5, 10.5, 5.0 Hz, 1H, CH=CHCH₂), 5.00 (d, J = 10.5 Hz, 1H, CO₂CH), 4.04 (t, J = 6.5 Hz, 1H, (CH₃)₂CCH(OTBS)), 3.92 (brs, 1H, CHOH(CHCH₃)), 3.05–3.01 (m, 2H, CH₃CH(C=O) and CHO-H(CHCH₃)), 2.78 (d, J = 6.5 Hz, 2H, CH₂COO), 2.68 (ddd, J = 14.0, 10.5, 10.5 Hz, 1H, CH=CHCH₂), 2.44 (s, 3H, CH₃Ar), 2.36–2.29 (m, 1H), 2.06 (dd, J = 14.5, 5.0 Hz, 1H, CH=CHCH₂), 1.98 (s, 3H, ArCH=CCH₃), 1.98–1.91 (m, 1H), 1.78–1.73 (m, 1H), 1.66–1.59 (m, 1H), 1.48–1.41 (m, 1H), 1.28–1.13 (m, 2H), 1.16 (s, 6H, C(CH₃)₂), 1.12 (d, 3H, J = 6.5 Hz, CH₃CH(C=O)), 1.00 (d, 3H, J = 7.0 Hz, CH₃CHCH₂), 0.81 (s, 9H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), 0.10 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), -0.07 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$); ^{13}C NMR (150.9 MHz, CDCl_3): δ = 218.0, 170.9, 160.8, 138.0, 137.6, 135.7, 134.7, 123.9, 115.8, 78.7, 76.3, 73.2, 53.5, 43.0, 39.0, 38.8, 33.5, 31.9, 28.4, 27.8, 26.1, 24.8, 22.9, 18.6, 16.5, 15.5, 14.1, 13.8, -3.6, -5.5; HRMS (FAB): calcd for $\text{C}_{32}\text{H}_{53}\text{CsNO}_6\text{Si}$ ($M + \text{Cs}^+$) 708.2697, found 708.2732.

13: R_f = 0.46 (silica gel, 30% EtOAc in hexanes); $[\alpha]_D^{22}$ = -40.0 (c = 0.67, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3465, 2932, 2873, 1741, 1693, 1586, 1096, 835 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.51 (s, 1H, ArH), 6.33 (s, 1H, ArCH=CCH₃), 5.35 (ddd, J = 14.5, 7.0, 7.0 Hz, 1H, CH=CHCH₂), 5.24 (ddd, J = 14.5, 7.0, 7.0 Hz, 1H, CH=CHCH₂), 5.17 (dd, J = 6.5, 3.5 Hz, 1H, CO₂CH), 4.41 (dd, J = 8.0, 3.5 Hz, 1H, (CH₃)₂CCH(OTBS)), 3.85 (brs, 1H, CHOH(CHCH₃)), 3.38 (brs, 1H, CHOH(CHCH₃)), 3.18 (qd, J = 7.0, 5.5 Hz, 1H, CH₃CH(C=O)), 2.65 (dd, J = 15.5, 8.0 Hz, 1H, CH₂COO), 2.60 (dd, J = 15.5, 3.5 Hz, 1H, CH₂COO), 2.54 (ddd, J = 14.5, 7.0, 3.5 Hz, 1H, CH=CHCH₂), 2.46–2.41 (m, 1H), 2.44 (s, 3H, CH₃Ar), 2.37 (ddd, J = 14.5, 7.0, 7.0 Hz, 1H, CH=CHCH₂), 2.19–2.11 (m, 1H), 1.96 (s, 3H, CH₃C=CH), 1.67–1.52 (m, 2H), 1.45 (dddd, J = 13.0, 13.0, 3.5, 3.5 Hz, 1H), 1.31–0.99 (m, 2H), 1.22 (d, 3H, J = 7.0 Hz, CH₃CH(C=O)), 1.14 (s, 3H, C(CH₃)₂), 1.09 (s, 3H, C(CH₃)₂), 1.02 (d, 3H, J = 7.0 Hz, CH₃CHCH₂), 0.84 (s, 9H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), 0.08 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$), -0.01 (s, 3H, $\text{Si}(\text{C}(\text{CH}_3)_3(\text{CH}_3)_2)$); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 218.3, 170.1, 160.9, 137.5, 136.3, 135.2, 134.6, 125.0, 115.8, 77.1, 75.1, 74.1, 54.0, 43.6, 40.7, 38.4, 35.3, 32.9, 30.9, 26.8, 26.1, 23.2, 21.8, 18.4, 16.8, 16.2, 14.6, 13.7, -3.9, -4.5; HRMS (FAB): calcd for $\text{C}_{32}\text{H}_{54}\text{NO}_6\text{Si}$ ($M + \text{H}^+$) 576.3720, found 576.3735.

***cis*-Dihydroxylactone 14—desilylation of compound 4:** Silyl ether 4 (55 mg, 0.096 mmol) was treated with a freshly prepared solution of 20% (v/v) trifluoroacetic acid (TFA)/ CH_2Cl_2 (9.6 mL, 0.01 M) at 0 °C. The reaction mixture was stirred at 0 °C for 30 min (completion of the reaction by TLC), and the solvents were evaporated under reduced pressure. The crude reaction mixture was purified by flash column chromatography (silica gel, ether) to obtain *cis*-dihydroxylactone 14 (39 mg, 89%). R_f = 0.21 (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22}$ = -46.5 (c = 0.71, CHCl_3); IR (thin film): $\tilde{\nu}_{\text{max}}$ = 3406, 2930, 1733, 1686, 1584, 1251, 733 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.47 (s, 1H, ArH), 6.31 (s, 1H, ArCH=C(CH₃)), 5.43 (ddd, J = 10.5, 10.5, 4.0 Hz, 1H, CH=CHCH₂), 5.36 (ddd, J = 10.5, 10.5, 4.5 Hz, 1H, CH=CHCH₂), 5.28 (d, J = 9.5 Hz, 1H, CO₂CH), 4.15 (d, J = 11.0 Hz, 1H, (CH₃)₂CCH(OH)), 3.72 (m, 1H, CHOH(CHCH₃)), 3.11 (qd, J = 7.0, 2.5 Hz, 1H, CH₃CH(C=O)), 3.02 (brs, 2H, OH), 2.66 (ddd, J = 15.0, 10.0, 10.0 Hz, 1H, CH=CHCH₂), 2.50 (dd, J = 15.5, 11.0 Hz, 1H, CH₂COO), 2.43 (s, 3H, CH₃Ar), 2.38 (dd, J = 15.5, 2.5 Hz, 1H, CH₂COO), 2.26–2.13 (m, 2H), 2.07–1.98 (m, 1H), 1.98 (s, 3H, ArCH=CCH₃), 1.80–1.60 (m,

2H), 1.37–1.13 (m, 3H), 1.31 (s, 3H, C(CH₃)₂), 1.17 (d, J = 7.0 Hz, 3H, CH₃CH(C=O)), 1.07 (s, 3H, C(CH₃)₂), 0.98 (d, J = 7.0 Hz, 3H, CH₃CHCH₂); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 220.3, 170.3, 160.9, 137.8, 137.4, 135.6, 133.5, 124.8, 115.9, 78.3, 74.2, 72.6, 53.0, 42.0, 39.0, 38.5, 32.4, 31.6, 27.6, 27.5, 22.5, 19.3, 15.9, 15.6, 13.8, 13.7; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_6$ ($M + \text{H}^+$) 462.2856, found 462.2844.

***trans*-Dihydroxylactone 15—desilylation of compound 13:** Silyl ether 13 (27 mg, 0.047 mmol) was treated with a freshly prepared solution of 20% (v/v) trifluoroacetic acid (TFA)/ CH_2Cl_2 (4.7 mL, 0.01 M) at 0 °C for 40 min, according to the procedure described for *cis*-dihydroxylactone 14, to yield, after flash column chromatography (silica gel, 50% EtOAc in hexanes), *trans*-dihydroxylactone 15 (29 mg, 95%). R_f = 0.22 (silica gel, ether); $[\alpha]_D^{22}$ = -37.9 (c = 0.70, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3400, 2933, 1733, 1688, 1583, 1466, 1251, 756 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.49 (s, 1H, ArH), 6.29 (s, 1H, ArCH=CCH₃), 5.50 (ddd, J = 15.0, 7.5, 7.5 Hz, 1H, CH=CHCH₂), 5.37 (dd, J = 5.5, 5.5 Hz, 1H, CO₂CH), 5.34 (ddd, J = 15.0, 7.5, 7.5 Hz, 1H, CH=CHCH₂), 4.19 (d, J = 10.0, 3.0 Hz, 1H, (CH₃)₂CCH(OH)), 3.73 (dd, J = 5.5, 2.5 Hz, 1H, CHOH(CHCH₃)), 3.26 (qd, J = 7.0, 5.5 Hz, 1H, CH₃CH(C=O)), 3.01 (brs, 1H, OH), 2.86 (brs, 1H, OH), 2.55 (dd, J = 15.5, 10.0 Hz, 1H, CH₂COO), 2.49 (dd, J = 15.5, 3.0 Hz, 1H, CH₂COO), 2.45 (s, 3H, CH₃Ar), 2.46–2.40 (m, 2H), 2.23–2.14 (m, 1H), 2.00–1.92 (m, 1H), 1.96 (s, 3H, ArCH=CCH₃), 1.64–1.56 (m, 2H), 1.47 (ddd, J = 12.5, 12.5, 4.0, 4.0 Hz, 1H), 1.40–1.00 (m, 2H), 1.26 (s, 3H, C(CH₃)₂), 1.18 (d, J = 7.0 Hz, 3H, CH₃CH(C=O)), 1.06 (s, 3H, C(CH₃)₂), 0.98 (d, J = 6.5 Hz, 3H, CH₃CHCH₂); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 219.9, 170.5, 161.0, 137.4, 136.8, 135.6, 134.5, 125.6, 116.1, 77.2, 76.1, 72.5, 52.3, 43.8, 38.8, 37.6, 36.1, 32.6, 30.3, 27.2, 21.5, 20.4, 16.6, 16.0, 15.1, 13.7; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_6$ ($M + \text{H}^+$) 462.2856, found 462.2866.

20-Oxa-epothilone A analogues 2 and 16—epoxidation of *cis*-dihydroxylactone

14: To a solution of *cis*-dihydroxylactone 14 (15.7 mg, 0.034 mmol) in methanol (170 μL , 0.2 M) was added acetonitrile (36 μL , 0.680 mmol, 20 equiv), potassium hydrogen carbonate (10.0 mg, 0.102 mmol, 3 equiv), and hydrogen peroxide (35 wt % solution in water; 33.0 μL , 0.374 mmol, 11 equiv), and the reaction mixture was stirred at ambient temperature for 3 h. Additional acetonitrile (36 μL , 0.680 mmol, 20 equiv), potassium hydrogen carbonate (10.0 mg, 0.102 mmol, 3 equiv), and hydrogen peroxide (35 wt % solution in water; 33.0 μL , 0.374 mmol, 11 equiv) were added, and the stirring was continued for 3 more hours, until a ca. 1:1 ratio of product(s) and starting material was indicated by TLC. The reaction mixture was then immediately passed through a short pad of silica gel with ether and concentrated. Purification by preparative thin-layer chromatography (250 μm silica gel plate, 50% EtOAc in hexanes) provided epoxide 2 (5.5 mg, 34%) and a trace amount of its α -isomeric epoxide 16 along with unreacted starting material 14 (5.2 mg, 33%).

2: R_f = 0.23 (silica gel, Ether); $[\alpha]_D^{25}$ = -25.2 (c = 0.31, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3417, 2927, 2866, 1731, 1692, 1584, 1260, 756 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.50 (s, 1H, ArH), 6.35 (s, 1H, ArCH=CCH₃), 5.44 (dd, 1H, J = 8.0, 2.5 Hz, CO₂CH), 4.12 (dd, 1H, J = 10.0, 3.0 Hz, (CH₃)₂CCH(OH)), 3.81 (dd, J = 5.0, 4.0 Hz, 1H, CHOH(CHCH₃)), 3.66 (brs, 1H, OH), 3.23 (qd, J = 7.0, 5.5 Hz, 1H, CH₃CH(C=O)), 3.02 (ddd, J = 7.0, 5.0, 5.0 Hz, 1H, CH₂CH-O(epoxide)CH), 2.90 (ddd, J = 7.0, 4.0, 4.0 Hz, 1H, CH₂CH-O(epoxide)CH), 2.54 (dd, J = 14.5, 10.0 Hz, 1H, CH₂COO), 2.46 (s, 3H, CH₃Ar), 2.45 (dd, J = 14.5, 3.0 Hz, 1H, CH₂COO), 2.08 (ddd, J = 15.0, 5.0, 3.0 Hz, 1H, CH₂CH-O(epoxide)CH), 2.01 (s, 3H, ArCH=CCH₃), 1.88 (ddd, J = 15.0, 7.5, 7.5 Hz, 1H, CH₂CH-O(epoxide)CH), 1.78–1.72 (m, 1H), 1.72–1.65 (m, 1H), 1.60–1.15 (m, 5H), 1.36 (s, 3H, C(CH₃)₂), 1.17 (d, 3H, J = 7.0 Hz, CH₃CH(C=O)), 1.11 (s, 3H, C(CH₃)₂), 1.00 (d, J = 7.0 Hz, 3H, CH₃CHCH₂); ^{13}C NMR (150.9 MHz, CDCl_3): δ = 220.6, 170.9, 161.3, 137.5, 136.8, 136.1, 116.5, 76.3, 74.8, 73.7, 57.4, 54.2, 52.4, 43.6, 38.5, 36.0, 31.1, 30.2, 26.7, 23.6, 21.1, 20.9, 17.0, 15.5, 14.1, 13.6; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_7$ ($M + \text{H}^+$) 478.2805, found 478.2790.

Larger amounts of epoxide 16 were synthesized by utilizing the method described for the epoxidation of *cis*-dihydroxylactone 39. **16:** R_f = 0.22 (silica gel, 2 × 50% EtOAc in hexanes); $[\alpha]_D^{25}$ = -22.0 (c = 0.10, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3486, 2926, 2856, 1735, 1689, 1460, 1256, 1148, 982 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.49 (s, 1H, ArH), 6.35 (s, 1H, ArCH=CCH₃), 5.68 (d, J = 8.5 Hz, 1H, CO₂CH), 4.11 (dd, J = 11.0, 2.0 Hz, 1H, (CH₃)₂CCH(OH)), 4.03–4.00 (m, 1H, CHOH(CHCH₃)), 3.86

CH_3CHCH_2 ; ^{13}C NMR (125.7 MHz, CDCl_3): δ = 221.9, 171.0, 160.9, 137.5, 136.5, 135.8, 134.3, 124.8, 116.0, 77.8, 74.6, 73.2, 52.7, 41.0, 38.2, 36.1, 34.5, 32.9, 32.8, 24.8, 23.0, 17.8, 16.0, 15.7, 13.8, 11.6; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_6$ ($M + \text{H}^+$) 462.2856, found 462.2843.

20-Oxa-epothilone A analogues 23 and 24—epoxidation of *cis*-dihydroxylactone 21: As described in the procedure for the epoxidation of *cis*-dihydroxylactone 14, acetonitrile (96 μL , 1.840 mmol, 40 equiv), potassium hydrogen carbonate (27.6 mg, 0.276 mmol, 6 equiv), and hydrogen peroxide (35 wt% solution in water; 90 μL , 1.012 mmol, 22 equiv) were added portionwise, over a period of 6 h, to a solution of *cis*-dihydroxylactone 21 (21.0 mg, 0.046 mmol) in methanol (230 μL , 0.2 M) to yield, after purification by preparative thin-layer chromatography (250 μm silica gel plate, 50% EtOAc in hexanes), epoxides 23 (or 24) (5.2 mg, 24%) and 24 (or 23) (2.0 mg, 9%) along with unreacted starting compound 21 (5.0 mg, 24%).

23 (or 24): R_f = 0.45 (silica gel, 75% EtOAc in hexanes); $[\alpha]_{\text{D}}^{22}$ = -25.6 (c = 0.25, CHCl_3); IR (thin film): $\tilde{\nu}_{\text{max}}$ = 3454, 2927, 1731, 1689, 1585, 1460, 1381, 1288, 1152, 1109, 1056, 979, 920 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.51 (s, 1H, ArH), 6.44 (s, 1H, ArCH=CCH₃), 5.62–5.58 (m, 1H, CO₂CH), 4.43 (d, J = 2.1 Hz, 1H, (CH₃)₂CCH(OH)), 4.30–4.26 (m, 1H, (CH₃)₂CCH(OH)), 3.82 (d, J = 8.5 Hz, 1H, CHOH(CHCH₃)), 3.35 (brs, 1H, CHOH(CHCH₃)), 3.21 (q, J = 7.0 Hz, 1H, CH₃CH(C=O)), 3.09 (ddd, J = 10.5, 4.0, 3.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.89 (ddd, J = 10.5, 4.5, 2.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.55–2.46 (m, 2H, CH₂COO), 2.45 (s, 3H, CH₃Ar), 2.17 (ddd, J = 15.5, 3.5, 3.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.07 (s, 3H, ArCH=CCH₃), 1.92 (ddd, J = 15.5, 10.5, 3.5 Hz, 1H, CH₂CH-O(epoxide)CH), 1.80–1.10 (m, 7H), 1.14 (s, 3H, C(CH₃)₂), 1.14 (d, J = 7.0 Hz, 3H, CH₃CH(C=O)), 1.06 (s, 3H, C(CH₃)₂), 1.02 (d, J = 7.0 Hz, 3H, CH₃CHCH₂); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 221.2, 171.7, 160.9, 137.6, 136.0, 134.2, 115.8, 75.8, 74.4, 72.8, 56.5, 53.9, 53.1, 40.2, 39.2, 34.2, 32.8, 29.6, 28.2, 22.9, 21.1, 16.4, 16.1, 15.2, 13.9, 12.0; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_7$ ($M + \text{H}^+$) 478.2805, found 478.2820.

24 (or 23): R_f = 0.42 (silica gel, 75% EtOAc in hexanes); $[\alpha]_{\text{D}}^{22}$ = -30.0 (c = 0.06, CHCl_3); IR (thin film): $\tilde{\nu}_{\text{max}}$ = 3416, 2924, 1733, 1687, 1585, 1459, 1380, 1149, 1068, 929 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.50 (s, 1H, ArH), 6.33 (s, 1H, ArCH=CCH₃), 5.87 (d, J = 10.5 Hz, 1H, CO₂CH), 4.73 (d, J = 9.0 Hz, 1H, (CH₃)₂CCH(OH)), 3.71 (brs, 1H, OH), 3.68 (d, J = 9.5 Hz, 1H, CHOH(CHCH₃)), 3.43 (brs, 1H, OH), 3.42 (q, J = 7.0 Hz, 1H, CH₃CH(C=O)), 3.19 (ddd, J = 10.0, 4.0, 2.0 Hz, 1H, CH₂CH-O(epoxide)CH), 3.08 (ddd, J = 9.0, 4.5, 4.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.51 (d, J = 14.0, 1H, CH₂COO), 2.45 (s, 3H, CH₃Ar), 2.45 (dd, J = 14.0, 9.0 Hz, 1H, CH₂COO), 2.13 (ddd, J = 14.5, 2.0, 2.0 Hz, 1H, CH₂CH-O(epoxide)CH), 2.03–1.97 (m, 1H, CH₂CH-O(epoxide)CH), 2.00 (s, 3H, ArCH=CCH₃), 1.80–0.96 (m, 7H), 1.16 (s, 3H, C(CH₃)₂), 1.13 (d, J = 7.0 Hz, 3H, CH₃CH(C=O)), 1.04 (s, 3H, C(CH₃)₂), 1.03 (d, J = 7.0 Hz, 3H, CH₃CHCH₂); HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{39}\text{NO}_7$ ($M + \text{Cs}^+$) 610.1781, found 610.1793.

20-Oxa-epothilone A analogues 25 and 26—epoxidation of *trans*-dihydroxylactone 22: As described in the procedure for the epoxidation of *cis*-dihydroxylactone 14, acetonitrile (62.6 μL , 1.200 mmol, 40 equiv), potassium hydrogen carbonate (18.0 mg, 0.180 mmol, 6 equiv), and hydrogen peroxide (35 wt% solution in water; 59 μL , 0.668 mmol, 22 equiv) were added portionwise, over a period of 6 h, to a solution of *trans*-dihydroxylactone 22 (14.0 mg, 0.030 mmol) in methanol (150 μL , 0.2 M) to yield, after purification by preparative thin-layer chromatography (250 μm silica gel plate, 50% EtOAc in hexanes), epoxides 25 (or 26) (2.2 mg, 15%) and 26 (or 25) (2.8 mg, 19%) along with unreacted starting material 22 (4.2 mg, 30%).

25 (or 26): R_f = 0.36 (silica gel, 50% EtOAc in hexanes); $[\alpha]_{\text{D}}^{22}$ = -23.5 (c = 0.55, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3445, 2931, 1731, 1583, 1460, 1374, 1287, 1214, 1151, 1108, 1064, 983 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.48 (s, 1H, ArH), 6.31 (s, 1H, ArCH=CCH₃), 5.71 (d, J = 10.0 Hz, 1H, CO₂CH), 4.39 (dd, J = 9.5, 4.0 Hz, 1H, (CH₃)₂CCH(OH)), 4.27 (d, J = 4.5 Hz, 1H, OH), 3.75 (d, J = 6.0 Hz, 1H, CHOH(CHCH₃)), 3.26 (q, J = 7.0 Hz, 1H, CH₃CH(C=O)), 2.92 (brs, 1H, OH), 2.91 (ddd, J = 9.0, 2.5, 2.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.71 (dd, J = 8.5, 2.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.50 (d, J = 13.0 Hz, 1H, CH₂COO), 2.44 (s, 3H, CH₃Ar), 2.39 (dd, J = 13.0, 10.0 Hz, 1H, CH₂COO), 2.13 (ddd, J = 15.0, 2.5, 2.5 Hz, 1H, CH₂CH-O(epoxide)CH), 1.98 (s, 3H, ArCH=CCH₃), 1.95–1.88 (m, 1H), 1.73–1.35 (m, 7H), 1.16 (s, 3H, C(CH₃)₂), 1.14 (d, 3H, J = 7.0 Hz, CH₃CH(C=O)), 1.03 (s, 3H, C(CH₃)₂), 0.93 (d, 3H, J = 7.0 Hz,

CH_3CHCH_2); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 220.7, 172.2, 161.0, 137.4, 136.4, 136.0, 116.6, 76.7, 74.6, 73.6, 58.0, 55.9, 53.0, 42.1, 39.1, 36.1, 34.6, 33.0, 29.2, 22.2, 21.7, 16.0, 15.3, 13.8, 13.5, 12.3.

26 (or 25): R_f = 0.35 (silica gel, 50% EtOAc in hexanes); $[\alpha]_{\text{D}}^{22}$ = -23.5 (c = 0.55, CHCl_3); IR (film): $\tilde{\nu}_{\text{max}}$ = 3418, 2930, 1733, 1686, 1584, 1460, 1376, 1152, 1049, 921 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ = 7.49 (s, 1H, ArH), 6.33 (s, 1H, ArCH=CCH₃), 5.54 (dd, J = 9.0, 3.0 Hz, 1H, CO₂CH), 4.20 (dd, J = 10.5, 1.5 Hz, 1H, (CH₃)₂CCH(OH)), 3.73 (d, J = 7.5 Hz, 1H, CHOH(CHCH₃)), 3.51 (brs, 1H), 3.35 (dq, J = 7.0, 1.5 Hz, 1H, CH₃CH(C=O)), 3.11 (brs, 1H, OH), 2.74 (dt, J = 5.5, 2.0 Hz, 2H, CH₂CH-O(epoxide)CH), 2.70 (dt, J = 5.5, 2.5 Hz, 1H, CH₂CH-O(epoxide)CH), 2.60 (dd, J = 15.5, 1.5 Hz, 1H, CH₂COO), 2.45 (s, 3H, CH₃Ar), 2.40 (dd, J = 15.5, 10.5 Hz, 1H, CH₂COO), 2.10 (ddd, J = 14.5, 5.5, 3.0 Hz, 1H, CH₂CH-O(epoxide)CH), 1.98 (s, 3H, ArCH=CCH₃), 1.91 (ddd, J = 14.5, 9.0, 5.5 Hz, 1H, CH₂CH-O(epoxide)CH), 1.68–1.25 (m, 7H), 1.29 (s, 3H, C(CH₃)₂), 1.12 (d, J = 7.0 Hz, 3H, CH₃CH(C=O)), 1.03 (s, 3H, C(CH₃)₂), 0.97 (d, J = 7.0 Hz, 3H, CH₃CHCH₂); ^{13}C NMR (125.7 MHz, CDCl_3): δ = 221.5, 171.4, 161.0, 137.4, 136.1, 135.7, 116.5, 76.7, 74.1, 72.8, 58.7, 55.7, 52.6, 42.4, 38.3, 35.2, 35.1, 32.9, 31.7, 22.6, 21.9, 18.5, 15.5, 15.0, 13.8, 12.1; HRMS (FAB): calcd for $\text{C}_{26}\text{H}_{40}\text{NO}_7$ ($M + \text{H}^+$) 478.2805, found 478.2823.

Spirocyclopropane ketoester 35—cyclopropanation of ethyl propionylacetate

34: Ethyl propionylacetate 34 (75.0 mL, 0.526 mol) was added to a solution of dry K₂CO₃ (218.0 g, 1.579 mol, 3.0 equiv) in DMF (526 mL, 1 M) at ambient temperature. This mixture was treated with 1,2-dibromoethane (60.0 mL, 0.684 mol, 1.3 equiv) over a period of 15 min and then rapidly stirred for 15 h, after which time completion of the reaction was indicated by NMR. Following filtration through Celite and washing with ether, the solvents were removed in vacuo. Vacuum distillation (b.p. 64 °C/6 mmHg) of the crude product resulted in the isolation of the pure spirocyclopropane ketoester 35 (53.9 g, 60%) as a colorless oil. R_f = 0.45 (silica gel, 17% EtOAc in hexanes); IR (film): $\tilde{\nu}_{\text{max}}$ = 2981, 2940, 1726, 1703, 1372, 1314, 1183, 1098 cm^{-1} ; ^1H NMR (250 MHz, CDCl_3): δ = 4.20 (q, J = 7.1 Hz, 2H, CH₃CH₂O), 2.86 (q, J = 7.3 Hz, 2H, CH₃CH₂), 1.43 (s, 4H, C(CH₃)₂), 1.29 (t, J = 7.1 Hz, 3H, CH₃CH₂O), 1.08 (t, J = 7.3 Hz, 3H, CH₃CH₂); ^{13}C NMR (62.9 MHz, CDCl_3): δ = 206.0, 171.1, 61.2, 35.1, 34.6, 18.5, 14.0, 8.3.

Spirocyclopropane ketoaldehyde 33—LiAlH₄ reduction/Swern oxidation of spirocyclopropane ketoester 35:

To a solution of spirocyclopropane ketoester 35 (53.9 g, 0.316 mol) in ether (1.5 L, 0.2 M) was added a solution of lithium aluminum hydride (LAH; 1 M solution in THF, 632 mL, 0.632 mol, 2.0 equiv) at -20°C over a period of 2 h, and the reaction mixture stirred at -20°C for 2 h. The reaction mixture was then diluted with ether (250 mL) and quenched by the sequential dropwise addition of water (24 mL), 15% aqueous sodium hydroxide solution (24 mL) and additional water (72 mL). The resulting slurry was allowed to warm to 25 °C over 10 h, and the aluminum salts were removed by filtration through Celite. The filtrate was dried (MgSO₄), and the solvent removed in vacuo to yield the crude diol (38.5 g, 93%), which was used in the oxidation step without further purification. An analytical sample was prepared by flash column chromatography (silica gel, 33 → 50% EtOAc in hexanes); R_f = 0.17 (silica gel, 50% EtOAc in hexanes); IR (film): $\tilde{\nu}_{\text{max}}$ = 3355, 2964, 2934, 2877, 1462, 1433, 1101, 1029, 969 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ = 4.11 (ddd, J = 11.3, 3.9, 1.3 Hz, 1H, CH₃CH₂CHOH), 3.03 (dd, J = 11.3, 5.9 Hz, 1H, CH₂OH), 2.97–2.85 (m, 3H, CH₂OH, CH₂OH and CHOH), 1.75–1.59 (m, 2H, CH₃CH₂), 0.97 (t, J = 7.5 Hz, 3H, CH₃CH₂), 0.61–0.53 (m, 2H, C(CH₃)₂), 0.43–0.36 (m, 2H, C(CH₃)₂); ^{13}C NMR (62.9 MHz, CDCl_3): δ = 80.1, 67.2, 27.3, 25.4, 10.7, 9.9, 7.9; HRMS (FAB): calcd for $\text{C}_7\text{H}_{14}\text{NaO}_2$ ($M + \text{Na}^+$) 153.0892, found 153.0894.

To a solution of oxalyl chloride (35.5 mL, 0.407 mol, 3.0 equiv) in CH₂Cl₂ (360 mL) was added dropwise DMSO (38.5 mL, 0.543 mol, 4.0 equiv) in CH₂Cl₂ (100 mL) at -78°C over 1 h. After the mixture had been stirred for 35 min, a solution of crude diol (17.7 g, 0.136 mol) in CH₂Cl₂ (200 mL) was added dropwise at -78°C over a period of 1.5 h. The solution was stirred for a further 1 h at -78°C , before Et₃N (151 mL, 1.085 mol, 8.0 equiv) was added over 40 min. After a further 15 min at -78°C the resulting slurry was allowed to warm to 0 °C over 1 h. Ether (700 mL) and saturated aqueous NH₄Cl solution (500 mL) were then added and the organic phase separated. The aqueous phase was again extracted with ether (500 mL), and the combined organic solution washed with saturated aqueous NH₄Cl solution (1.0 L), dried (Na₂SO₄), filtered, and concentrated under reduced pressure.

Purification by flash column chromatography (silica gel, 25% ether in hexanes) afforded spirocyclopropane ketoaldehyde **33** (10.9 g, 64%). $R_f = 0.57$ (silica gel, 50% EtOAc in hexanes); (b.p. 45 °C/1.5 mm Hg); IR (film): $\tilde{\nu}_{\max} = 2974, 2939, 1723, 1699, 1318, 1176, 1099 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 9.83$ (s, 1H, CHO), 2.70 (q, $J = 7.2$ Hz, 2H, CH_2CH_2), 1.70–1.68 (m, 2H, $\text{C}(\text{CH}_2)_2$), 1.57–1.54 (m, 2H, $\text{C}(\text{CH}_2)_2$), 1.09 (t, $J = 7.2$ Hz, 3H, CH_3CH_2); $^{13}\text{C NMR}$ (62.9 MHz, CDCl_3): $\delta = 205.9, 197.8, 40.9, 33.7, 21.2, 7.7$; HRMS (FAB): calcd for $\text{C}_7\text{H}_{11}\text{O}_2$ ($M + \text{H}^+$) 127.0759, found 127.0765.

Silyl ether 32—allylboration of spirocyclopropane ketoaldehyde 33 and silylation: Allylmagnesium bromide (1 M solution in ether, 80 mL, 80.0 mmol, 1.0 equiv) was added dropwise to a well-stirred solution of (–)-Ipc₂BOME (27.2 g, 86.0 mmol, 1.1 equiv) in ether (500 mL) at 0 °C. After the completion of the addition, the gray slurry was stirred at room temperature for 1 h, and the solvent removed under reduced pressure. Pentane (400 mL) was added to the residual solids, and the mixture stirred for 10 min. The stirring was discontinued to allow precipitation of the magnesium salts, and the clear supernatant pentane solution was transferred through a cannula carefully avoiding the transfer of any solid materials. This process was repeated four times. The combined pentane fractions were concentrated to a volume of ca. 500 mL and then added dropwise, without further purification, to a solution of ketoaldehyde **33** (10.1 g, 79.7 mmol, 1.0 equiv) in ether (250 mL) at –100 °C. After the addition was complete, the mixture was stirred at the same temperature for 30 min. Methanol (10 mL) was added at –100 °C, and the reaction mixture was allowed to warm to –40 °C over 40 min. Saturated aqueous NaHCO_3 solution (125 mL), followed by H_2O_2 (50 wt% solution in H_2O , 50 mL) were added, and the reaction mixture was allowed to stir at room temperature for 12 h. The organic phase was separated, and the aqueous phase extracted with EtOAc (3 × 250 mL). The combined organic extracts were washed with saturated aqueous NH_4Cl solution (500 mL), dried (Na_2SO_4), and the solvents removed in vacuo to yield the crude allylic alcohol, which was used without further purification. An analytical sample was prepared by flash column chromatography (silica gel, 3% acetone in CH_2Cl_2); $R_f = 0.14$ (silica gel, 17% EtOAc in hexanes); $[\alpha]_D^{22} = -93.6$ ($c = 0.92$, CHCl_3), IR (film): $\tilde{\nu}_{\max} = 3472, 2978, 2938, 1678, 1641, 1376, 1068, 994, 972, 914 \text{ cm}^{-1}$; $^1\text{H NMR}$ (500 MHz, CDCl_3): $\delta = 5.84$ (dddd, $J = 17.0, 10.0, 7.0, 7.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.10–5.03 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 3.30 (dd, $J = 6.5, 6.5$ Hz, 1H, CHOH), 3.18 (brs, 1H, CHOH), 2.42–2.37 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 2.12 (q, $J = 7.5$ Hz, 2H, CH_3CH_2), 1.26 (ddd, $J = 9.5, 7.0, 5.0$ Hz, 1H, $\text{C}(\text{CH}_2)_2$), 1.16 (ddd, $J = 9.5, 7.0, 5.0$ Hz, 1H, $\text{C}(\text{CH}_2)_2$), 1.07 (ddd, $J = 9.0, 7.0, 5.0$ Hz, 1H, $\text{C}(\text{CH}_2)_2$), 1.00 (t, $J = 7.5$ Hz, 3H, CH_3CH_2), 0.94 (ddd, $J = 9.0, 7.0, 5.0$ Hz, 1H, $\text{C}(\text{CH}_2)_2$); $^{13}\text{C NMR}$ (62.9 MHz, CDCl_3): $\delta = 212.5, 135.4, 117.0, 74.9, 39.7, 35.4, 29.4, 14.0, 12.6, 8.0$; HRMS (FAB): calcd for $\text{C}_{10}\text{H}_{16}\text{NaO}_2$ ($M + \text{Na}^+$) 191.1048, found 191.1042.

This crude alcohol was dissolved in CH_2Cl_2 (750 mL, 0.3 M), and the solution was cooled to –78 °C. The solution was treated with 2,6-lutidine (40 mL, 0.368 mol, 4.6 equiv), and after stirring for 5 min, *tert*-butyldimethylsilyl triflate (70 mL, 0.305 mmol, 3.8 equiv) was added dropwise. The reaction mixture was allowed to stir at –78 °C for 35 min, after which time no starting material was detected by TLC. Saturated aqueous NH_4Cl solution (500 mL) was added, and the reaction mixture was allowed to warm to room temperature. The organic phase was separated, and the aqueous layer extracted with ether (3 × 300 mL). The combined organic extracts were dried (MgSO_4) and filtered through Celite, and the solvents were removed in vacuo to yield the crude silyl ether **32**, which was used without further purification. An analytical sample was prepared by flash column chromatography (silica gel, 2 → 17% ether in hexanes); $R_f = 0.50$ (silica gel, 17% EtOAc in hexanes); $[\alpha]_D^{22} = +20.3$ ($c = 0.94$, CHCl_3), IR (film): $\tilde{\nu}_{\max} = 2955, 2932, 2890, 2857, 1687, 1256, 1086, 838, 776 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 5.79$ (dddd, $J = 17.0, 10.0, 7.0, 7.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.01–4.94 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.16 (dd, $J = 5.5, 5.5$ Hz, 1H, CHOTBS), 2.38–2.21 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 2.30 (q, $J = 7.5$ Hz, 2H, CH_3CH_2), 1.13–1.09 (m, 1H, $\text{C}(\text{CH}_2)_2$), 1.00 (t, $J = 7.5$ Hz, 3H, CH_3CH_2), 0.98–0.90 (m, 3H, $\text{C}(\text{CH}_2)_2$), 0.86 (s, 9H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.04 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.01 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$); $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): $\delta = 210.5, 135.5, 116.8, 70.2, 41.6, 36.3, 31.0, 25.8, 18.0, 12.6, 11.3, 8.3, -4.3, -4.6$; HRMS (FAB): calcd for $\text{C}_{16}\text{H}_{31}\text{O}_2\text{Si}$ ($M + \text{H}^+$) 283.2093, found 283.2087.

Spirocyclopropane ketoacid 31—oxidation of olefin 32: The crude alkene **32** was dissolved in MeCN (143 mL), CCl_4 (143 mL), and H_2O (214 mL), and

the mixture cooled to 0 °C. Sodium periodate (70 g, 327 mmol, 4.1 equiv) and ruthenium(III) chloride hydrate (898 mg, 3.98 mmol, 0.05 equiv) were added, and the mixture was stirred at 0 °C for 10 min. The dark mixture was allowed to warm to ambient temperature and stirred for 3 h, after which time the disappearance of starting material was indicated by TLC. CH_2Cl_2 (1.5 L) and saturated aqueous NaCl solution (1.5 L) were added, and the layers were separated. Extractions of the aqueous phase with CH_2Cl_2 (3 × 750 mL), filtration through Celite, concentration and flash column chromatography (2 → 80% EtOAc in hexanes) yielded pure spirocyclopropane ketoacid **31** (10.2 g, 43% for three steps); $R_f = 0.39$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22} = -0.8$ ($c = 1.19$, CHCl_3); IR (film): $\tilde{\nu}_{\max} = 2955, 2930, 2857, 1712, 1687, 1255, 1090, 838, 778 \text{ cm}^{-1}$; $^1\text{H NMR}$ (500 MHz, CDCl_3): $\delta = 4.45$ (dd, $J = 5.5, 5.5$ Hz, 1H, CHOTBS), 2.62 (dd, $J = 15.5, 5.5$ Hz, 1H, CH_2CHOTBS), 2.61 (dd, $J = 15.5, 5.5$ Hz, 1H, CH_2CHOTBS), 2.39 (dq, $J = 17.5, 7.0$ Hz, 1H, CH_3CH_2), 2.28 (dq, $J = 17.5, 7.0$ Hz, 1H, CH_3CH_2), 1.01 (t, $J = 7.0$ Hz, 3H, CH_3CH_2), 0.84 (s, 9H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.06 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.04 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$); $^{13}\text{C NMR}$ (150.9 MHz, CDCl_3): $\delta = 210.2, 177.4, 68.7, 42.2, 36.4, 30.9, 25.7, 18.0, 13.0, 12.6, 8.2, -4.6, -4.9$; HRMS (FAB): calcd for $\text{C}_{15}\text{H}_{28}\text{NaO}_4\text{Si}$ ($M + \text{Na}^+$) 323.1655, found 323.1650.

Hydroxyacids 29 and 36—aldol condensation of ketoacid 31 with aldehyde 8:

In accordance with the procedure described for the preparation of aldols **6** and **11**, ketoacid **31** (1.581 g, 5.26 mmol) in THF (18 mL) was treated with lithium diisopropylamide [LDA; freshly prepared from *n*BuLi (7.73 mL, 1.6 M solution in hexanes, 12.37 mmol, 2.35 equiv) and diisopropylamine (1.70 mL, 12.10 mmol, 2.3 equiv) in THF (53 mL)] and aldehyde **8** (1.13 g, 8.94 mmol, 1.7 equiv) in THF (53 mL) to afford a mixture of aldol products **29** and **36** in a ratio of ca. 2:3 ($^1\text{H NMR}$). This crude material was used without further purification.

Esters 28 and 37—EDC coupling of alcohol 30 with the mixture of ketoacids 29 and 36:

By analogy to the procedure described above for the synthesis of esters **5** and **12**, a solution of the mixture of ketoacids **29** and **36** (2.289 g crude), 4-DMAP (66 mg, 0.540 mmol), and alcohol **30** (2.81 g, 13.43 mmol) in CH_2Cl_2 (8.0 mL) was treated with EDC (1.23 g, 6.42 mmol) to provide, after column chromatography (silica gel, 33 → 50% ether in hexanes), ester **28** (488 mg, 15% from ketoacid **31**) and ester **37** (1.171 g, 36% from ketoacid **31**).

28: $R_f = 0.38$ (silica gel, 50% ether in hexanes); IR (film): $\tilde{\nu}_{\max} = 3508, 3078, 2926, 2855, 1737, 1675, 1378, 1255, 1170, 1095, 987, 836 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.96$ (s, 1H, ArH), 6.50 (s, 1H, $\text{ArCH}=\text{CCH}_3$), 5.87–5.65 (m, 2H, $2\text{CH}_2\text{CH}=\text{CH}_2$), 5.28 (dd, $J = 6.8, 6.8$ Hz, 1H, CO_2CH), 5.10 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.04 (d, $J = 10.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.00 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.93 (d, $J = 10.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.30 (dd, $J = 6.2, 5.0$ Hz, 1H, $(\text{CH}_2)_2\text{CCH}(\text{OTBS})$), 3.48 (brs, 1H, $\text{CHOH}(\text{CHCH}_3)$), 3.42 (d, $J = 9.2$ Hz, 1H, $\text{CHOH}(\text{CHCH}_3)$), 2.98 (q, $J = 6.5$ Hz, 1H, $\text{CH}_3\text{CH}(\text{C}=\text{O})$), 2.70 (s, 3H, CH_3Ar), 2.66 (dd, $J = 15.0, 6.8$ Hz, 1H, CH_2COO), 2.56 (dd, $J = 15.0, 5.0$ Hz, 1H, CH_2COO), 2.51–2.45 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 2.09–2.02 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 2.06 (d, $J = 1.0$ Hz, 3H, $\text{ArCH}=\text{CCH}_3$), 1.78–1.74 (m, 1H), 1.73–1.63 (m, 1H), 1.63–1.48 (m, 2H), 1.34–0.96 (m, 5H), 1.01 (d, $J = 7.0$ Hz, 3H, $\text{CH}_3\text{CH}(\text{C}=\text{O})$), 0.87 (s, 9H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.84 (d, $J = 7.0$ Hz, 3H, CH_3CHCH_2), 0.08 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$), 0.07 (s, 3H, $\text{Si}(\text{CH}_3)_3(\text{CH}_3)_2$); $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): $\delta = 216.0, 170.2, 164.5, 152.3, 138.9, 136.6, 133.2, 121.1, 117.7, 116.3, 114.0, 78.6, 74.3, 69.2, 42.6, 40.5, 37.4, 35.9, 35.4, 34.1, 32.3, 26.0, 25.6, 19.1, 17.9, 15.3, 14.6, 14.0, 12.5, 10.0, -4.5, -5.0$; HRMS (FAB): calcd for $\text{C}_{34}\text{H}_{55}\text{CsNO}_5\text{SSi}$ ($M + \text{Cs}^+$) 750.2625, found 750.2649.

37: $R_f = 0.30$ (silica gel, 50% ether in hexanes); $[\alpha]_D^{22} = -12.7$ ($c = 1.38$, CHCl_3); IR (film): $\tilde{\nu}_{\max} = 3499, 3077, 2931, 2857, 1738, 1674, 1375, 1254, 1169, 1096, 982, 836 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.96$ (s, 1H, ArH), 6.50 (s, 1H, $\text{ArCH}=\text{CCH}_3$), 5.83–5.69 (m, 2H, $2\text{CH}_2\text{CH}=\text{CH}_2$), 5.29 (dd, $J = 6.8, 6.8$ Hz, 1H, CO_2CH), 5.10 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.04 (d, $J = 10.5$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 5.00 (d, $J = 17.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.95 (d, $J = 10.0$ Hz, 1H, $\text{CH}_2\text{CH}=\text{CH}_2$), 4.32 (dd, $J = 6.5, 4.8$ Hz, 1H, $(\text{CH}_2)_2\text{CCH}(\text{OTBS})$), 3.50–3.46 (m, 1H, $\text{CHOH}(\text{CHCH}_3)$), 3.24 (brd, $J = 2.0$ Hz, 1H, $\text{CHOH}(\text{CHCH}_3)$), 2.94 (qd, $J = 7.0, 2.5$ Hz, 1H, $\text{CH}_3\text{CH}(\text{C}=\text{O})$), 2.70 (s, 3H, CH_3Ar), 2.65 (dd, $J = 15.1, 6.5$ Hz, 1H, CH_2COO), 2.56 (dd, $J = 15.1, 4.8$ Hz, 1H, CH_2COO), 2.51–2.45 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 2.06 (d, $J = 1.0$ Hz, 3H, $\text{ArCH}=\text{CCH}_3$),

4,4-Ethano-epothilone A analogues 48 and 49—epoxidation of *cis*-dihydroxylactone 46: As described in the procedure for the epoxidation of *cis*-dihydroxylactone **39**, *cis*-hydroxylactone **46** (11.0 mg, 0.023 mmol) in MeCN (200 µL) and CH₂Cl₂ (300 µL) was treated with a 0.4 mM aqueous solution of Na₂EDTA (120 µL), 1,1,1-trifluoroacetone (200 µL), Oxone® (114 mg, 0.185 mmol, 8.0 equiv), and NaHCO₃ (25 mg, 0.296 mmol, 12.8 equiv), to yield, after purification by preparative thin-layer chromatography (250 µm silica gel plate, 17% acetone in CH₂Cl₂), epoxides **48** (or **49**) (4.0 mg, 39%) and **49** (or **48**) (4.5 mg, 35%).

48 (or **49**): $R_f = 0.30$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22} = -92.1$ ($c = 0.14$, CHCl₃); IR (thin film): $\tilde{\nu}_{max} = 3468, 2922, 2854, 1735, 1668, 1456, 1378, 1257, 1161, 1093, 980, 733\text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.97$ (s, 1H, ArH), 6.54 (s, 1H, ArCH=C(CH₃)), 5.78 (dd, $J = 5.0, 4.5$ Hz, 1H, CO₂CH), 4.12 (brd, $J = 11.0$ Hz, 1H, (CH₂)₂CCHOH), 3.73 (q, $J = 7.0$ Hz, 1H, CH₃CH(C=O)), 3.62 (d, $J = 10.0$ Hz, 1H, CHO-H(CHCH₃)), 3.59 (brs, 1H, OH), 3.51–3.47 (m, 1H, OH), 3.12 (ddd, $J = 6.5, 6.5, 4.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 3.00 (ddd, $J = 6.5, 6.5, 4.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.72 (dd, $J = 16.0, 11.0$ Hz, 1H, CH₂COO), 2.70 (s, 3H, CH₃Ar), 2.55 (dd, $J = 16.0, 2.5$ Hz, 1H, CH₂COO), 2.11 (d, $J = 1.0$ Hz, 3H, ArCH=CCH₃), 2.00–1.93 (m, 2H), 1.75–1.06 (m, 8H), 1.13 (d, $J = 7.0$ Hz, 3H, CH₃CH(C=O)), 1.06 (d, $J = 6.5$ Hz, 3H, CH₃CHCH₂), 1.06–0.95 (m, 2H, C(CH₂)₂), 0.71–0.68 (m, 1H, C(CH₂)₂); ¹³C NMR (150.9 MHz, CDCl₃): $\delta = 218.4, 171.1, 160.5, 152.0, 136.2, 119.7, 116.5, 75.7, 73.7, 70.1, 56.6, 54.8, 43.3, 39.5, 34.2, 32.4, 31.9, 31.0, 29.7, 26.5, 21.4, 19.2, 15.8, 15.7, 11.3, 11.2$; HRMS (FAB): calcd for C₂₆H₃₈NO₆S ($M + H^+$) 492.2420, found 492.2434.

49 (or **48**): $R_f = 0.30$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22} = -98.0$ ($c = 0.21$, CHCl₃); IR (thin film): $\tilde{\nu}_{max} = 3460, 2923, 2855, 1736, 1669, 1454, 1378, 1251, 1159, 1040, 977, 733\text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.97$ (s, 1H, ArH), 6.59 (s, 1H, ArCH=C(CH₃)), 5.62 (dd, $J = 6.0, 3.5$ Hz, 1H, CO₂CH), 4.26 (d, $J = 10.5$ Hz, 1H, (CH₂)₂CCHOH), 3.87 (brs, 1H, OH), 3.71 (d, $J = 9.2$ Hz, 1H, CHO-H(CHCH₃)), 3.64–3.56 (m, 2H, CH₃CH(C=O) and OH), 3.09 (ddd, $J = 6.0, 6.0, 4.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.97 (ddd, $J = 6.0, 6.0, 4.5$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.70 (s, 3H, CH₃Ar), 2.62 (dd, $J = 15.5, 10.5$ Hz, 1H, CH₂COO), 2.41 (dd, $J = 15.5, 2.5$ Hz, 1H, CH₂COO), 2.13 (s, 3H, ArCH=CCH₃), 2.08–1.97 (m, 2H), 1.75–1.03 (m, 8H), 1.09 (d, $J = 7.0$ Hz, 3H, CH₃CH(C=O)), 1.07 (dd, $J = 6.5$ Hz, 3H, CH₃CHCH₂), 1.00–0.90 (m, 2H, C(CH₂)₂), 0.74 (ddd, $J = 9.5, 7.0, 4.5$ Hz, 1H, C(CH₂)₂); ¹³C NMR (150.9 MHz, CDCl₃): $\delta = 217.7, 170.8, 164.9, 152.1, 135.6, 119.9, 116.7, 76.4, 73.6, 69.8, 56.6, 54.8, 42.6, 39.7, 34.6, 32.7, 30.9, 30.5, 29.7, 28.2, 21.2, 19.2, 17.1, 16.1, 15.8, 11.2$; HRMS (FAB): calcd for C₂₆H₃₈NO₆S ($M + H^+$) 492.2420, found 492.2431.

4,4-Ethano-epothilone A analogues 50 and 51—epoxidation of *trans*-dihydroxylactone 47: As described in the procedure for the epoxidation of *cis*-dihydroxylactone **39**, *trans*-hydroxylactone **47** (20 mg, 0.042 mmol) in MeCN (400 µL) and CH₂Cl₂ (600 µL) was treated with a 0.4 mM aqueous solution of Na₂EDTA (400 µL), 1,1,1-trifluoroacetone (250 µL), Oxone® (207 mg, 0.334 mmol, 8.0 equiv), and NaHCO₃ (45 mg, 0.538 mmol, 12.8 equiv), to yield, after purification by preparative thin-layer chromatography (250 µm silica gel plate, 50% EtOAc in hexanes), epoxides **50** (or **51**) (4.5 mg, 22%) and **51** (or **50**) (5.6 mg, 27%).

50 (or **51**): $R_f = 0.20$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22} = -49.5$ ($c = 0.33$, CHCl₃); IR (thin film): $\tilde{\nu}_{max} = 3472, 2923, 2855, 1734, 1666, 1457, 1374, 1263, 1163, 1089, 981, 910, 731\text{ cm}^{-1}$; ¹H NMR (500 MHz, CDCl₃): $\delta = 6.99$ (s, 1H, ArH), 6.56 (s, 1H, ArCH=C(CH₃)), 5.49 (dd, $J = 9.0, 2.5$ Hz, 1H, CO₂CH), 4.28–4.25 (m, 1H, (CH₂)₂CCHOH), 3.87 (d, $J = 3.5$ Hz, 1H, OH), 3.69 (qd, $J = 7.0, 2.0$ Hz, 1H, CH₃CH(C=O)), 3.63 (d, $J = 8.5$ Hz, 1H, CHO-H(CHCH₃)), 3.59 (brs, 1H, OH), 2.89 (ddd, $J = 5.5, 5.5, 2.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.78 (ddd, $J = 5.5, 5.5, 2.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.71 (s, 3H, CH₃Ar), 2.63 (dd, $J = 15.5, 10.5$ Hz, 1H, CH₂COO), 2.51 (dd, $J = 15.5, 2.5$ Hz, 1H, CH₂COO), 2.15 (ddd, $J = 15.0, 9.0, 6.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.09 (d, $J = 1$ Hz, 3H, ArCH=CCH₃), 1.95 (ddd, $J = 15.0, 5.0, 2.5$ Hz, 1H, CH₂CH-O(epoxide)CH), 1.66–1.20 (m, 8H), 1.16 (d, $J = 7.0$ Hz, 3H, CH₃CH(C=O)), 1.03 (d, $J = 6.5$ Hz, 3H, CH₃CHCH₂), 1.02–0.96 (m, 2H, C(CH₂)₂), 0.65–0.61 (m, 1H, C(CH₂)₂); ¹³C NMR (150.9 MHz, CDCl₃): $\delta = 217.5, 171.4, 165.0, 152.0, 136.3, 119.9, 116.6, 76.4, 74.5, 69.5, 57.7, 55.5, 44.0, 39.7, 34.9, 34.6, 34.9, 32.7, 30.6, 29.7, 22.5, 19.2, 16.2, 15.3, 11.6, 10.6$; HRMS (FAB): calcd for C₂₆H₃₇CsNO₆S ($M + Cs^+$) 624.1396, found 624.1421.

51 (or **50**): $R_f = 0.15$ (silica gel, 50% EtOAc in hexanes); $[\alpha]_D^{22} = -73.5$ ($c = 0.14$, CHCl₃); IR (thin film): $\tilde{\nu}_{max} = 3463, 2917, 2852, 1735, 1668, 1456, 1377, 1259, 1160, 1095, 910, 734\text{ cm}^{-1}$; ¹H NMR (500 MHz, CDCl₃): $\delta = 6.97$ (s, 1H, ArH), 6.55 (s, 1H, ArCH=C(CH₃)), 5.63 (dd, $J = 11.5, 2.5$ Hz, 1H, CO₂CH), 3.81 (brs, 1H, OH), 3.73 (q, $J = 7.0$ Hz, 1H, CH₃CH(C=O)), 3.69 (d, $J = 11.0$ Hz, 1H, (CH₂)₂CCHOH), 3.66 (d, $J = 9.0$ Hz, 1H, CHO-H(CHCH₃)), 3.38 (brs, 1H, OH), 2.79 (dd, $J = 17.0, 11.0$ Hz, 1H, CH₂COO), 2.73–2.69 (m, 1H, CH₂CH-O(epoxide)CH), 2.70 (s, 3H, CH₃Ar), 2.68–2.64 (m, 2H, CH₂CH-O(epoxide)CH and CH₂COO), 2.24 (ddd, $J = 14.5, 2.5, 2.5$ Hz, 1H, CH₂CH-O(epoxide)CH), 2.08 (d, $J = 1$ Hz, 3H, ArCH=CCH₃), 2.06–1.99 (m, 1H), 1.93–1.85 (m, 1H), 1.77 (ddd, $J = 14.5, 11.5, 8.0$ Hz, 1H, CH₂CH-O(epoxide)CH), 1.72–1.68 (m, 1H), 1.67–1.08 (m, 5H), 1.53 (ddd, $J = 9.5, 7.0, 4.5$ Hz, 1H, C(CH₂)₂), 1.15 (d, $J = 7.0$ Hz, 3H, CH₃CH(C=O)), 1.09 (d, $J = 7.0$ Hz, 3H, CH₃CHCH₂), 0.94 (ddd, $J = 9.5, 7.0, 4.5$ Hz, 1H, C(CH₂)₂), 0.59 (ddd, $J = 9.5, 6.5, 4.5$ Hz, 1H, C(CH₂)₂); ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 217.9, 171.3, 164.9, 151.9, 136.2, 120.6, 116.8, 76.7, 73.2, 71.4, 60.0, 57.2, 44.3, 39.3, 35.9, 34.7, 34.2, 32.5, 32.3, 29.6, 21.5, 19.1, 18.8, 16.1, 14.7, 11.1, 11.1$; HRMS (FAB): calcd for C₂₆H₃₇CsNO₆S ($M + Cs^+$) 624.1396, found 624.1431.

Molecular dynamics and minimization calculations (CV Force Field) were performed on a SGI Indigo-2 workstation using Insight II (Biosym Technologies, Inc., San Diego, CA). Pictures were created using AVS (AVS Inc., Waltham, MA) and locally developed modules running on a DEC Alpha 3000/500 with a Kubota Pacific Denali graphics card (we thank Chris Boddy and Stefan Stefan for their assistance in these modeling studies).

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